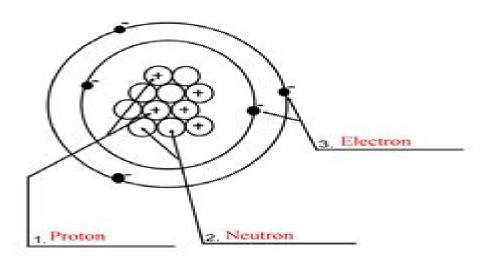
#### Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



- What type of charge does a proton have? Positive (+1)
- What type of charge does a neutron have? Neutral (0)
- What type of charge does an electron have? Negative (-1)
- Which two subatomic particles are located in the nucleus of an atom?
   Protons and neutrons
- 8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
  35
- 9. What is the atomic number of the atom in the diagram above?
- What is the atomic mass/mass number of the atom in the diagram above?
   (11+11=22) \*protons and neutrons added together) 22 is the atomic mass/mass number
- 11. How many protons are in the nucleus of an atom with an atomic number of 15?
  15 \*number of protons is the same as the atomic number
- How many electrons are in the nucleus of an atom with an atomic number of 20?
- How many neutrons are in the nucleus of an atom with an atomic number of 25? (use Periodic Table for mass)
   Neutrons, atomic mass: 54.938 ~ 55 atomic number: 25....55-25=30
- 14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
  Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.
- How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
   Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

## **Chapter Atomic Structure Worksheet Answers**

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Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 **Exercises for Principles of Imaging Science and Protection** Michael A. Thompson, Steven B. Dowd, Janice D. Hall, Marian P. Hattaway, 1994 A smart collection of additional review questions and practise problems to further reinforce basic concepts The laboratory manual section includes about 30 basic experiments which concentrate more on radiographic factors which affect the quality of the resulting radiograph Readers with a weaker math science background will appreciate the remediation sections and self tests conclude each section to gauge the reader s weaknesses **Educart CBSE Class 9** Science One-shot Question Bank 2026 (Strictly for 2025-26 Exam) Educart, 2025-06-07 What Do You Get Question Bank for daily practiceHandpicked important chapter wise questions What notable components are included in Educart CBSE CLASS 9 Science ONE SHOT Chapter wise concept mapsEach chapter has 3 worksheets for daily practiceUnit wise worksheets Pull Out are given separately for extra practiceNCERT Exemplar DIKSHA PYQs Competency Based Important Qs to cover every type of questions Answer key for every worksheetDetailed explanation of each question with Related Theory Caution Important PointsPYQs from annual papers of various schoolsStrictly based on 28th March 2025 CBSE syllabus Why choose this book The Educart CBSE Class 9 Science One Shot book helps students master concepts quickly with visual concept maps and daily practice worksheets It builds exam confidence through targeted Qs from NCERT Exemplar DIKSHA and PYQs With detailed explanations and syllabus alignment it ensures smart effective preparation for scoring higher in Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science exams ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Learning Elementary Chemistry for Class 8 (A.Y. 2023-24)Onward Dr. R. Goel, 2023-05-20 The series Learning Elementary Chemistry for Classes 6 to 8 has been revised strictly according to the latest curriculum The content of this series has been developed to fulfill the requirement of all the six domains Concepts Processes Applications Attitudes Creativity and World view of Science to make teaching and learning of Chemistry interesting understandable and enjoyable for young minds This series builds a solid foundation for young learners to prepare them for higher classes The main strength of the series lies in the subject matter and the experience that a learner will get in solving difficult and complex problems of Chemistry Emphasis has been laid upon mastering the fundamental principles of Chemistry rather than specific procedures Unique features of this series are The content of the book is written in a very simple and easy to understand language All the Key concepts in the curriculum have been systematically covered and graded in the text Each theme has been divided into units followed by thought provoking and

engaging exercises to test the knowledge understanding and applications of the concepts learnt in that unit At the end of each theme a comprehensive theme assignment which is aligned with the guidelines provided in National Education Policy NEP 2020 is given Explanations illustrations diagrams experiments and solutions to numerical problems have been included to make the subject more interesting comprehensive and appealing Diagrams illustrations and text have been integrated to enhance comprehension Definitions and other important scientific information are highlighted Throughout the series investigations related to the text enable the learners to learn through experimentation Quick revision of each chapter has been given under the caption Highlights in Review Online Support It provides Video lectures Unit wise interactive exercises Chapterwise Worksheet Solution of textbook questions for Teachers only E Book for Teachers only I hope this series would meet the needs and requirements of the curriculum to achieve the learning outcomes as laid down in the curriculum Suggestions and constructive feedback for the further improvement of the book shall be gratefully acknowledged and incorporated in the future edition of the book Author

Earth Science Samuel N. Namowitz, Nancy E. Spaulding, 1989

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Simplified ICSE Chemistry Viraf J. Dalal, Science Interactions ,1996 Turbo Pascal Programming Gary B. Shelly, Thomas J. Cashman, Steven G. Forsythe, 1987 This programming textbook emphasizes file processing early and features step by step explanations of sound programming design and structured programming in a problem orientated approach There is extensive coverage of single and multiple dimensional arrays arrays of records and sequential and binary array searching techniques Procedures recursion and sorting are comprehensively discussed and every Pascal statement required in each program example is explained and illustrated Numerous problems are included illustrating typical business

scientific and general interest applications End of chapter review questions coding exercises structured work through exercises and student programming assignments are also present This book should be of interest to degree and diploma students on introductory courses in Pascal programming **Building Block** Prentice-Hall Staff,1994 *Holt Science and Technology 2002* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2002 *Te HS&T J* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004-02 **Science Interactions** Robert W. Avakian,1995-07-17 *Te HS&T a* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004-02 <u>Te HS&T 2007 Shrt Crs M</u> Holt Rinehart & Winston,2007

#### Chapter Atomic Structure Worksheet Answers Book Review: Unveiling the Magic of Language

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