

# Chapter 10 Chemical Quantities Review

## Part 1: Vocabulary

Label the following descriptions with the correct vocabulary word.

- |   |   |
|---|---|
| _____ mass  | _____ percent composition.                                |
| _____ empirical formula   | _____ empirical formula                                   |
| _____ molar mass  | _____ standard temperature and pressure (STP, 1 atm, 0°C) |
| _____ 10 <sup>23</sup>  |   |
| _____ is a description of the relative amounts of each element in a compound _____ percent composition                    |   |
| _____ the lowest whole number ratio of the atoms of the elements in a compound _____ empirical formula                    |   |
| _____ the volume of 1 mole of any gas at STP _____ 22.4 L   |   |
| _____ the smallest particle of an element or a compound _____ molar mass  |   |
| _____ the number corresponding to 6.02 x 10 <sup>23</sup> representative particles of a substance _____ Avogadro's number |   |
| _____ 1 mole = 6.02 <sup>23</sup> particles _____ empirical formula   |   |
| _____ the number corresponding to 6.02 x 10 <sup>23</sup> particles of any substance or element at STP _____ 22.4 L       |   |
| _____ 6.02 x 10 <sup>23</sup> representative particles (6.02 <sup>23</sup> is just)                                       |   |

## Part 2: Section Concept Review

### 10.1 The Mole: A Measurement

**Completion** Use the completion strategy to strengthen knowledge of the terms and your understanding of the concepts involved with the terms of the definitions completed with a term (don't guess) or explain.

Measuring whole units of something, or mass, and its volume for a single quantity called the \_\_\_\_\_ mole. \_\_\_\_\_ Representative representative particles like large or a substance is \_\_\_\_\_ Avogadro's number \_\_\_\_\_

It is the mass of \_\_\_\_\_ of a compound, elements and together the \_\_\_\_\_ percent mass \_\_\_\_\_ of the atoms making the compound. Other use techniques the substance the atoms (the whole) the \_\_\_\_\_ molar mass \_\_\_\_\_ of the compound. These are \_\_\_\_\_ 6.02 x 10<sup>23</sup> \_\_\_\_\_ representative particles the mass of one substance.

#### True or False

\_\_\_\_\_ 1 mole of any substance is 6.02 x 10<sup>23</sup> particles.

\_\_\_\_\_ 1 mole of substance contains 6.02 x 10<sup>23</sup> atoms.

\_\_\_\_\_ 1 mole of the representative particles of a substance is the molar mass. **True**

\_\_\_\_\_ 1 mole of 100 g is composed of one mole of carbon and four atoms of oxygen.

\_\_\_\_\_ 1 mole mass of nitrogen gas (N<sub>2</sub>) is **28.02 g** **28.02 g**

#### Problems

# Chapter Chemical Quantities Guided Reading Answer Key

**Ralph Thomas Myers**



## Chapter Chemical Quantities Guided Reading Answer Key:

*Introduction to Chemical Processes: Principles, Analysis, Synthesis* Regina M. Murphy, 2007 Introduction to Chemical Processes Principles Analysis Synthesis enhances student understanding of the connection between the chemistry and the process Users will find strong coverage of chemistry gain a solid understanding of what chemical processes do convert raw materials into useful products using energy and other resources and learn about the ways in which chemical engineers make decisions and balance constraints to come up with new processes and products The author presents material and energy balances as tools to achieve a real goal workable economical and safe chemical processes and products Loaded with intriguing pedagogy this text is essential to a students first course in Chemical Engineering Additional resources intended to guide users are also available as package options such as ChemSkill Builder

Understanding Chemistry: Chemical systems: thermochemistry, kinetics, and colligative properties, 1967 **Teaching Science for Understanding** James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities **Chemistry:**

**The Central Science** Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Darius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation **Holt Chemistry** Ralph Thomas Myers, 2004 **Understanding Chemistry: Chemical reactions**, 1967 **Technical Book Review Index**, 1971 *Economics* J. Holton Wilson, 1997

**Scientific American**, 1872 Monthly magazine devoted to topics of general scientific interest **Teacher's Wraparound Edition: Two Biology Everyday Experience** Albert Kaskel, 1994-04-19 *Prentice Hall Science* Anthea Maton, Jill D. Wright, Jean Hopkins, Susan Johnson, David Lahart, Maryanna Quon Warner, 1994 Engineering, 1962

**Bulletin of the Atomic Scientists**, 1972-10 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin's iconic Doomsday Clock stimulates solutions for a safer world **Backpacker**, 2007-09 Backpacker brings the outdoors straight to the reader's doorstep inspiring and enabling them to go more places and enjoy nature more often The authority on active adventure Backpacker is the world's first GPS enabled magazine and the only magazine whose editors personally test

the hiking trails camping gear and survival tips they publish Backpacker s Editors Choice Awards an industry honor recognizing design feature and product innovation has become the gold standard against which all other outdoor industry awards are measured      **Bulletin of the Atomic Scientists** ,1969-03 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin s iconic Doomsday Clock stimulates solutions for a safer world      *Bulletin of the Atomic Scientists* ,1958-01 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin s iconic Doomsday Clock stimulates solutions for a safer world      Webster's Third International Dictionary of the English Language, Unabridged ,1961

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