

## Honors Chemistry- Chapter 9 Homework Packet Stoichiometry

- 1) For each of the following conversion factors we use in stoichiometry, fill in the blanks with the quantities that each relates.

Mole Ratio: relates \_\_\_\_\_ to \_\_\_\_\_

Avogadro's #: relates \_\_\_\_\_ to \_\_\_\_\_

Molar Mass: relates \_\_\_\_\_ to \_\_\_\_\_

Molarity: relates \_\_\_\_\_ to \_\_\_\_\_

- 2) Write all the possible mole ratios for the following chemical reaction.



- 3) An aqueous solution is prepared by dissolving 74.5 grams of KCl in 300 mL of water. What is the molarity of the solution?

# Chapter 9 Stoichiometry Packet

**Ying Liu**



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