

Oxoacids (O—H bond): contain H, O, and a central non-metal  
 Acidity determined by the polarity of the O—H bond:

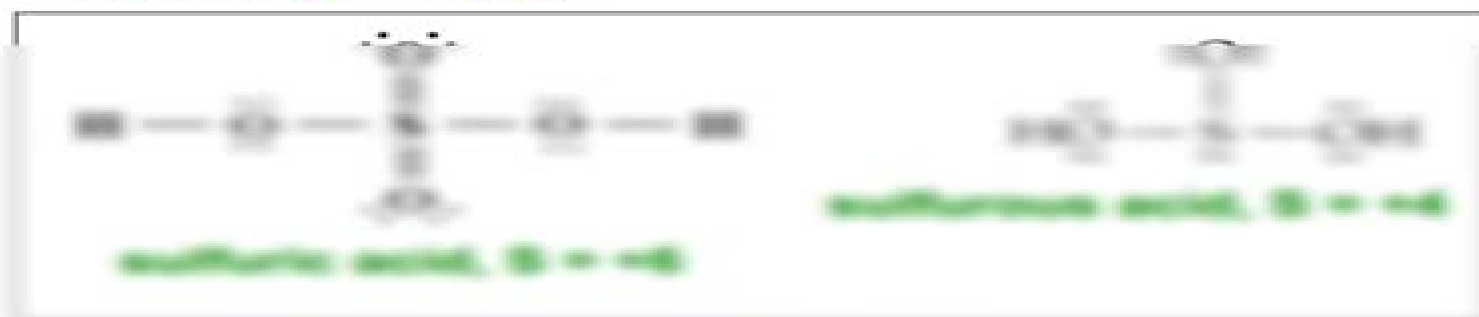
1. central atom from the same group of the periodic table, same oxidation number

Relative strength increases with increased electronegativity of the central atom.



2. same central atom, different oxidation number (# of O's)

Higher oxidation number = stronger acid.



# Chapter 8 3 Solution Acids And Bases Abswer Key

**Karen Timberlake**



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