

CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron (II) sulfate, the roman numeral tells us (2)
 - How many atoms of Fe are in one formula unit.
 - How many sulfate ions can be attached to the iron atom.
 - The charge on each Fe ion.
 - The total positive charge of the formula unit.
- The result of changing a subscript in a correctly written chemical formula is to (4)
 - Change the number of moles represented in the formula.
 - Change the charges on the other ions in the compound.
 - Change the formula so that it no longer represents the compound.
 - Have no effect on the formula.
- Derive a generalization for when an acid name will end in the suffix -ic or -ous. If ending in -ic, the anion ended in -ate. If ending in -ous, the anion ended in -ite.
- Derive a generalization for when an acid name will begin with the prefix hydro-. If prefix is hydro-, the anion ended in -ite. This is almost always a monatomic ion.
- Fill in the blanks in the table below.

Compound Name	Formula
Acetic acid	$\text{HC}_2\text{H}_3\text{O}_2$
Aluminum sulfate	$\text{Al}_2(\text{SO}_4)_3$
Lead II chloride	PbCl_2
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Hydrochloric acid	HCl
Dinitrogen pentoxide	N_2O_5
Copper II carbonate	CuCO_3
Sodium sulfate	Na_2SO_4
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Barium sulfate	BaSO_4
Nitrous acid	HNO_2
Magnesium perchlorate	$\text{Mg}(\text{ClO}_4)_2$
Iron II nitrate	$\text{Fe}(\text{NO}_3)_2$
Iron III nitrate	$\text{Fe}(\text{NO}_3)_3$
Sulfur trioxide	SO_3
Sulfur dioxide	SO_2
Barium perchlorate	$\text{Ba}(\text{ClO}_4)_2$
Tetraphosphorus trisulfide	P_4S_3
Phosphoric acid	H_3PO_4

- Name the following compound and identify its molecular formula (acid)
 - $\text{Ca}(\text{HCO}_3)_2$ calcium bicarbonate - II
 - $\text{Fe}(\text{NO}_3)_3$ iron III nitrate - II
 - $\text{Fe}(\text{ClO}_4)_3$ iron III perchlorate - II
 - N_2O_5 dinitrogen pentoxide - III

Chapter 9 Chemical Names And Formulas Test B Answers

Kenneth W. Whitten, Raymond E. Davis



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Ebook: Physical Science Tillery, 2016-04-16 Ebook Physical Science *General, Organic and Biochemistry* Kate Brown, William Henry Brown, 1983 **Meyer Brothers Druggist**, 1908 *Student's Guide to Introduction to Chemical Principles* by Edward I. Peters, 2d Ed Peter P. Berlow, 1978 *Chemistry* Charles H. Corwin, 1994 The book focuses on the concepts of chemistry and the applications that maintain and generate motivation for the subject of chemistry 10 in One Study Package for CBSE Science Class 10 with 3 Sample Papers & 16 Chapter Tests ebook Disha Experts, 2017-09-01 These books contain Access Codes along with instructions to access the Online Material In case you face any difficulty write to us at ebooks support aietco in 10 in ONE CBSE Study Package Science class 10 with 3 Sample Papers is another innovative initiative from Disha Publication This book provides the excellent approach to Master the subject The book has 10 key ingredients that will help you achieve success 1 Chapter Utility Score CUS 2 Exhaustive Theory with Concept Maps 3 Text Book exercises 4 VSA SA LA Questions 5 Past year questions including 2017 Solved papers 6 HOTS Value based Exemplar 7 Past NTSE Exemplar MCQs 8 16 Chapter Tests ebooks 9 Important Formulas Terms Definitions 10 3 Sample Papers with detailed solutions (Free Sample) 10 in One Study Package for CBSE Chemistry Class 12 with Objective Questions & 3 Sample Papers 4th Edition Disha Experts, 2021-07-21 *Chemistry* James E. Brady, John R. Holum, 1993-01-21 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping students develop skills in analytical thinking and problem solving Students are taught in a variety of ways to think of skills as tools that can be used to solve complex problems Several aids are included to help focus and inspire student interest frequent reference to common chemicals in commercial products numerous photographs of reactions in chapter practice exercises following worked examples *General Chemistry* Kenneth W. Whitten, Raymond E. Davis, 1996 Are you looking for the key to success in your chemistry class In CHEMISTRY you will find a strong molecular reasoning focus problem solving exercises and an innovative online homework management system that will prepare you for any challenge you might encounter The textbook is filled with learning aids that will help you master concepts of the course *Study Guide/Selected Solutions Manual* Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book *Basic Concepts of Chemistry* Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical

related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance

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