

Chemistry: Chapter 9 Structure and properties of substances

Combined Science (Chemistry Part): Chapter 9 Structure and properties of substances

Multiple-choice questions

Sections 9.1–9.2

[1]CMEMF0020900001[1] [1]CSEMF0020900001[1]

Chlorine exists as a gas at room temperature and pressure because

- A. chlorine is reactive.
- B. the intermolecular forces among chlorine molecules are weak.
- C. the intermolecular forces between chlorine atoms are weak.
- D. chlorine molecules are diatomic.



##B##

[1]CMEMF0020900002[1] [1]CSEMF0020900002[1]

Helium has a low boiling point because

- A. a helium atom has a duplet structure.
- B. helium molecules are monoatomic.
- C. the van der Waals' forces between helium atoms are weak.
- D. helium is a noble gas.



##C##

[1]CMEMF0020900003[1] [1]CSEMF0020900003[1]

Which of the following statements concerning the structure of solid carbon dioxide is/are correct?

- (1) It has a simple molecular structure.
 - (2) Carbon atom and oxygen atom within the molecule are held by ionic bond.
 - (3) Carbon dioxide molecules are held together by intermolecular forces.
- A. (1) only
 - B. (2) only
 - C. (1) and (3) only
 - D. (2) and (3) only



##C Carbon dioxide is a covalent molecule. There are two C=O bonds joining the carbon and oxygen atoms together in the molecule.##

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Derek Stebbens



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The Go-To Guide for Engineering Curricula, Grades 9-12 Cary I. Sneider, 2014-12-05 How to engineer change in your high school science classroom With the Next Generation Science Standards your students won't just be scientists they'll be engineers But you don't need to reinvent the wheel Seamlessly weave engineering and technology concepts into your high school math and science lessons with this collection of time tested engineering curricula for science classrooms Features include A handy table that leads you straight to the chapters you need In depth commentaries and illustrative examples A vivid picture of each curriculum its learning goals and how it addresses the NGSS More information on the integration of engineering and technology into high school science education

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