

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

- 1.** C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____.
 - (a) how many atoms of Fe are in one formula unit
 - (b) how many sulfate ions can be attached to the iron atom
 - (c) the charge on each Fe ion
 - (d) the total positive charge of the formula unit
- 2.** C The result of changing a subscript in a correctly written chemical formula is to _____.
 - (a) change the number of moles represented by the formula
 - (b) change the charges on the other ions in the compound
 - (c) change the formula so that it no longer represents the compound it previously represented
 - (d) have no effect on the formula
- 3.** The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.

| | |
|----------------------------------------------|--------------------------------------------------------------------------------------------------------|
| <u>4 elements</u> | a. How many elements make up this compound? |
| <u>6 oxygen atoms</u> | b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$? |
| <u>21 atoms</u> | c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$? |
| <u>4.2×10^{24} atoms</u> | d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$? |
- 4.** How many atoms are present in each of these formula units?

| | |
|-----------------|--------------------------------|
| <u>11 atoms</u> | a. $Ca(HCO_3)_2$ |
| <u>45 atoms</u> | b. $C_{12}H_{22}O_{11}$ |
| <u>10 atoms</u> | c. $Fe(ClO_2)_3$ |
| <u>9 atoms</u> | d. $Fe(ClO_3)_2$ |
- 5.**

| | |
|-----------------------------|----------------------------------------------------------------------|
| <u>N_2O_5</u> | a. What is the formula for the compound dinitrogen pentoxide? |
| <u>carbon(IV) sulfide</u> | b. What is the Stock name for the covalent compound CS_2 ? |
| <u>H_2SO_3</u> | c. What is the formula for sulfurous acid? |
| <u>phosphoric acid</u> | d. What is the name for the acid H_3PO_4 ? |

Chapter Chemical Names And Formulas Practice Problems Answer Key

Peter Atkins, Loretta Jones



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Understanding Chemistry Fred M. Dewey,1994 **Addison-Wesley Chemistry** ,1990 **Chemistry** John S. Phillips,Cheryl Wistrom,2000 **Introduction to Chemical Principles** Howard Stephen Stoker,1990 Introduction to Chemical Principles is a text for students who have had little to no previous instruction in chemistry or who had such instruction long enough ago that a thorough review is needed preface **Prentice Hall Chemistry** ,2000 *Introductory Chemistry* Darrell D. Ebbing,Albert E. Richardson,1995 **Community/Public Health Nursing Practice** Frances A. Maurer,Claudia M. Smith,2012-10-01 Focusing on practical need to know information Community Public Health Nursing Practice helps you learn how to apply the nursing process at the community and family level It features an engaging easy to understand writing style as well as assessment tools detailed case studies and clinical examples that demonstrate how key concepts apply to real world practice Additional resources on the companion Evolve website expand and enhance content within the text Practical features including Case Studies Ethics in Practice and The Nursing Process in Practice illustrate real world applications of key community public health nursing concepts A complete unit on the community as client helps you understand how the assessment diagnosis planning intervention and evaluation steps of the nursing process apply to the community as opposed to an individual A chapter devoted to community assessment provides a complete assessment tool and shows you how the tool applies to two different types of communities UNIQUE A chapter on screening and referral promotes population focused practice which is the crux of community public health nursing A separate unit on the family emphasizes the importance of viewing the family as a singular client A complete discussion of the Minnesota Wheel helps you better understand this widely accepted framework for community public health nursing practice Helpful sections such as Focus Questions Chapter Outlines Key Ideas and Learning by Experience and Reflection help you pinpoint essential information NEW Healthy People 2020 objectives throughout the text help you identify common health risk factors in populations and families NEW Coverage of health care reform including the Patient Protection and Affordable Care Act of 2010 PPACA explores how current health care legislation impacts community public health nursing NEW Discussions of community health hot button issues such as human trafficking genital circumcision and bullying introduce you to today s health care challenges NEW Information on weather related disaster fatalities bioterrorism and national and state planning responses familiarize you with current relevant issues which affect the health of populations worldwide and shape the role of the community public health nurse *Merrill Chemistry* Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 *Chemical Principles* Peter Atkins,Loretta Jones,2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry s frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in

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Preparatory Chemistry Howard Stephen Stoker, 1990 **Investigating Chemistry** Matthew Johll, 2006-03-17 Matthew Johll's book introduces students from a non science background to the fundamentals of chemistry through an array of examples and applications from real life crime scenes Sherlock Holmes stories and authentic accounts of drug deals murders and thefts.

Basic Concepts of Chemistry Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book. It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features. Each section focuses on the development and assessment of one or two specific objectives. Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives. These features are followed by a set of independent practice problems. The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics. Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance.

Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change *The Pharmaceutical Era*, 1891

Organic Chemistry T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2023 Organic Chemistry 13th edition provides a comprehensive yet accessible treatment of all the essential organic chemistry concepts with emphasis on relationship between structure and reactivity in the subject. The textbook includes all the concepts covered in a typical organic chemistry textbook but is unique in its skill development approach to the subject. Numerous hands on activities and real world examples are integrated throughout the text to help students understand both the why and the how behind organic chemistry. This International Adaptation offers new and updated content with improved presentation of all course material. It offers new material on several topics including the relevance of intermolecular forces in the immune response and vaccines like those for Covid 19 the chemistry of breathing carbonic anhydrase how conjugation and complexation affect the color of lobsters and how biodegradable polymers are used to stabilize vaccines and pharmaceuticals. Content is revised to reflect the current understanding of chemical processes and improved depictions of longstanding mechanisms. This edition builds on the ongoing pedagogical strength of the book with the inclusion of additional worked and end of chapter problems and an engaging set of new problems entitled Chemical Consultant Needed. These draw from the primary chemical literature.

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