

Electrons in Atoms

Section 5.1 Light and Quantized Energy

In your textbook, read about the wave nature of light.

Use each of the terms below just once to complete the passage.

amplitude
light

energy
wave

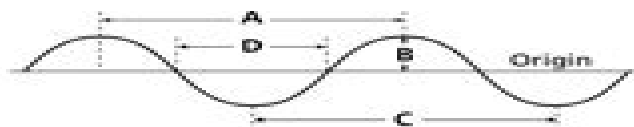
frequency
wavelength

hertz
speed

Electromagnetic radiation is a kind of **(1)** _____ that behaves like a(n) **(2)** _____ as it travels through space. **(3)** _____ is one type of electromagnetic radiation. Other examples include X rays, radio waves, and microwaves.

All waves can be characterized by their wavelength, amplitude, frequency, and **(4)** _____. The shortest distance between equivalent points on a continuous wave is called a(n) **(5)** _____. The height of a wave from the origin to a crest or from the origin to a trough is the **(6)** _____. **(7)** _____ is the number of waves that pass a given point in one second. The SI unit for frequency is the **(8)** _____, which is equivalent to one wave per second.

Use the figure to answer the following questions.



9. Which letter(s) represent one wavelength?

10. Which letter(s) represent the amplitude? _____
11. If twice the length of A passes a stationary point every second, what is the frequency of the wave?

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Robert S. Boikess



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Organic Chemistry, 5e Student Study Guide and Solutions Manual David R. Klein, Laurie S. Starkey, 2025-03-04 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry

Physics, Volume Two: Chapters 18-32 John D. Cutnell, Kenneth W. Johnson, 2014-12-15 Cutnell and Johnson has been the 1 text in the algebra based physics market for almost 20 years The 10th edition brings on new co authors David Young and Shane Stadler both out of LSU The Cutnell offering now includes enhanced features and functionality The authors have been extensively involved in the creation and adaptation of valuable resources for the text This edition includes chapters 18 32

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Organic Chemistry K. Peter C. Vollhardt, Neil E. Schore, 2014-01-01 With authors who are both accomplished researchers and educators Vollhardt and Schore's Organic Chemistry takes a functional group approach with a heavy emphasis on understanding how the structure of a molecule determines how that molecule will function in chemical reactions By understanding the connection between structure and function students will be better prepared to understand mechanisms and solve practical problems in organic chemistry The new edition brings in the latest research breakthroughs and applications expanded problem solving help and new online homework options

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Study Guide/Selected Solutions Manual Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book **A Textbook of Modern Physics** Le Roy Dougherty Weld, Frederic Palmer, 1925 Visualizing Microbiology Rodney P. Anderson, Linda Young, Kim R. Finer, 2020-12-16 The second edition of Visualizing Microbiology contains a completely redesigned TOC and the most current coverage of the COVID 19 pandemic This text is ideal for introductory microbiology courses for non majors and pre allied health students Visualizing Microbiology brings the narrative to life with an applied clinical focus helping students see and understand the unseen in the world of microbiology The unique visual pedagogy of the text provides a powerful combination of content and visuals ideal for microbiology **Matter, Electricity, Energy** Walther Gerlach, 1928 **... What is the Atom?** Edward Neville da Costa Andrade, 1927 Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry

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