

Hydrogen

- (C) Hydrated silicates of aluminium and sodium
(D) Silicates of calcium and magnesium
91. The approximate mass of tritium oxide molecule is
(A) 18 amu (B) 20 amu
(C) 22 amu (D) 24 amu
92. Heavy water is called as such because it
(A) is a dimer than common water
(B) is an oxide of protium
(C) has a heavy or bad taste
(D) has a heavier isotopes of hydrogen
93. When sulphur trioxide is heated with heavy water the products are
(A) Deutero-sulphuric acid
(B) Deuterium sulphurous acid
(C) Deuterium and sulphuric acid
(D) None of the above.
94. High dipole moment of water (1.03 D) justifies that
(A) it is not linear molecule
(B) it is a universal solvent
(C) it has higher density than ice
(D) it is neutral towards litmus.
95. Which of the following substances does not produce ammonia on reaction with water?
(A) CaCN_2 (B) AlN
(C) $\text{Ca}(\text{CN})_2$ (D) Mg_3N_2 .
96. In which of the following reactions does water act as oxidising agent?
(A) $3\text{F}_2 + 3\text{H}_2\text{O} \longrightarrow 6\text{HF} + \text{O}_3$
(B) $\text{C}(\text{g}) + \text{H}_2\text{O}(\text{g}) \longrightarrow \text{CO}(\text{g}) + \text{H}_2(\text{g})$
(C) $\text{Ca}_3\text{P}_2 + 6\text{H}_2\text{O} \longrightarrow 3\text{Ca}(\text{OH})_2 + 2\text{PH}_3$
(D) $2\text{Cl}_2 + 2\text{H}_2\text{O} \longrightarrow 4\text{HCl} + \text{Cl}_2$.
97. The formula of sodium zeolite which is used in permutit process for softening water is
(A) $\text{Na}_2\text{O} \cdot \text{Al}_2\text{O}_3 \cdot \text{SiO}_2 \cdot x\text{H}_2\text{O}$
(B) $\text{Na}_2\text{Al}_2\text{Si}_2\text{O}_4 \cdot x\text{H}_2\text{O}$
(C) $\text{Na}_2\text{O} \cdot \text{AlO}_3 \cdot \text{SiO}_4 \cdot x\text{H}_2\text{O}$
(D) $\text{K}_2\text{Al}_2\text{SiO}_8 \cdot x\text{H}_2\text{O}$.
98. Calgon causes the softening of hard water by
(A) sequestration of Ca^{2+} and Mg^{2+} ion
(B) sequestration of Cl^- and SO_4^{2-} ion
(C) precipitating the Ca^{2+} and Mg^{2+} ions as phosphates
(D) precipitating the Ca^{2+} and Mg^{2+} ions as sulphates.
99. The compound sodium polymetaphosphate ($\text{Na}[\text{Na}_4(\text{PO}_3)_6]$) is called calgon because
(A) It was developed by the scientist named Challaghan
(B) It was developed first in California
(C) It refers to calcium zone
(D) It is based on the name of the company which developed it.
100. Which of the following operations would cause removal of temporary hardness of water?
(A) passing CO_2 through it
(B) passing SO_2 through it
(C) adding calculated amount of $\text{Ca}(\text{OH})_2$
(D) adding calculated amount of sodium hypophosphate
101. A mixture of which pair of species react with water to produce a pure colourless gas that gives white fumes with HCl ?
(A) Calcium hydride, calcium carbide
(B) Calcium carbide and aluminium nitride
(C) Magnesium nitride and calcium nitride
(D) Calcium phosphide and calcium cyanamide
102. The maximum tolerable limit of degree of hardness in water required for our daily needs is approximately
(A) 10 to 15 ppm (B) 100 to 150 ppm
(C) 10^3 ppm (D) 700 to 800 ppm.
103. Of the two solvents H_2O and D_2O sodium chloride dissolves
(A) equally in both (B) more in D_2O
(C) more in H_2O (D) only in H_2O .
104. Mass percentage of deuterium in heavy water is
(A) same as that protium in water
(B) 20
(C) 11.1
(D) unpredictable.

Hydrogen peroxide

105. Hydrogen peroxide is used as
(A) Oxidising agent
(B) Reducing agent
(C) Both as oxidising and reducing agent
(D) Drying agent
106. In the laboratory, H_2O_2 is prepared by the action of
(A) cold H_2SO_4 on BaO_2
(B) aqueous alkali on Na_2O_2

Answers

91. (C) 92. (D) 93. (A) 94. (C) 95. (C) 96. (C) 97. (A) 98. (A) 99. (C) 100. (C)
101. (C) 102. (B) 103. (C) 104. (B) 105. (C) 106. (A)

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Arshad Iqbal



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