

CHAPTER 15 REVIEW

Acid-Base Titration and pH**SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. Calculate the following values without using a calculator.

- _____ a. The $[\text{H}_3\text{O}^+]$ is 1×10^{-6} M in a solution. Calculate the $[\text{OH}^-]$.
- _____ b. The $[\text{H}_3\text{O}^+]$ is 1×10^{-9} M in a solution. Calculate the $[\text{OH}^-]$.
- _____ c. The $[\text{OH}^-]$ is 1×10^{-12} M in a solution. Calculate the $[\text{H}_3\text{O}^+]$.
- _____ d. The $[\text{OH}^-]$ in part c is reduced by half, to 0.5×10^{-12} M. Calculate the $[\text{H}_3\text{O}^+]$.
- _____ e. The $[\text{H}_3\text{O}^+]$ and $[\text{OH}^-]$ are _____ (directly, inversely, or not) proportional in any system involving water.

2. Calculate the following values without using a calculator.

- _____ a. The pH of a solution is 2.0. Calculate the pOH.
- _____ b. The pOH of a solution is 4.73. Calculate the pH.
- _____ c. The $[\text{H}_3\text{O}^+]$ in a solution is 1×10^{-3} M. Calculate the pH.
- _____ d. The pOH of a solution is 5.0. Calculate the $[\text{OH}^-]$.
- _____ e. The pH of a solution is 1.0. Calculate the $[\text{OH}^-]$.

3. Calculate the following values.

- _____ a. The $[\text{H}_3\text{O}^+]$ is 2.34×10^{-5} M in a solution. Calculate the pH.
- _____ b. The pOH of a solution is 3.5. Calculate the $[\text{OH}^-]$.
- _____ c. The $[\text{H}_3\text{O}^+]$ is 4.6×10^{-8} M in a solution. Calculate the $[\text{OH}^-]$.

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.4. $[\text{H}_3\text{O}^+]$ in an aqueous solution = 2.3×10^{-3} M.

- _____ a. Calculate $[\text{OH}^-]$ in this solution.

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James C. Hill



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Chemistry All-in-One For Dummies (+ Chapter Quizzes Online) Christopher R. Hren, John T. Moore, Peter J. Mikulecky, 2022-11-23 Everything you need to crush chemistry with confidence Chemistry All in One For Dummies arms you with all the no nonsense how to content you'll need to pass your chemistry class with flying colors You'll find tons of practical examples and practice problems and you'll get access to an online quiz for every chapter Reinforce the concepts you learn in the classroom and beef up your understanding of all the chemistry topics covered in the standard curriculum Prepping for the AP Chemistry exam Dummies has your back with plenty of review before test day With clear definitions concise explanations and plenty of helpful information on everything from matter and molecules to moles and measurements Chemistry All in One For Dummies is a one stop resource for chem students of all valences Review all the topics covered in a full year high school chemistry course or one semester of college chemistry Understand atoms molecules and the periodic table of elements Master chemical equations solutions and states of matter Complete practice problems and end of chapter quizzes online Chemistry All In One For Dummies is perfect for students who need help with coursework or want to cram extra hard to ace that chem test **A Problem-Solving Approach to Aquatic Chemistry** James N. Jensen, 2023-01-05 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter's main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text A Problem Solving Approach to Aquatic Chemistry includes Fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria Solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques Acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity Complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry Oxidation reduction equilibria including equilibrium calculations graphical approaches and applications Gas liquid and solid liquid equilibrium with

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