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46. In three of the survey sites, parties were placed in the middle of the path. The paths in 28° and 32° of the mountain were different from the other two paths. In the southern path in 28° and 32° of the mountain, the distance to the adjacent slopes. The paths in the middle of the

- (a) 10^{-10} m
 (b) 10^{-11} m
 (c) 10^{-12} m
 (d) 10^{-13} m
10. The ratio of the energy of the n^{th} orbit of a hydrogen atom to the ground state of hydrogen is $\frac{1}{4}$. What will be the radius of the n^{th} orbit?
- (a) $0.25a_0$ (b) $0.5a_0$ (c) $0.75a_0$ (d) a_0
11. The ratio of the energy of the n^{th} orbit of a hydrogen atom to the ground state of hydrogen is $\frac{1}{9}$. What will be the radius of the n^{th} orbit?
- (a) $0.33a_0$ (b) $0.66a_0$ (c) $0.99a_0$ (d) $1.32a_0$
12. The energy of the n^{th} orbit of a hydrogen atom is -13.6 eV . What will be the radius of the n^{th} orbit?
- (a) $0.5a_0$ (b) $1.0a_0$ (c) $1.5a_0$ (d) $2.0a_0$
13. The energy of the n^{th} orbit of a hydrogen atom is -13.6 eV . What will be the radius of the n^{th} orbit?
- (a) $0.5a_0$ (b) $1.0a_0$ (c) $1.5a_0$ (d) $2.0a_0$
14. The energy of the n^{th} orbit of a hydrogen atom is -13.6 eV . What will be the radius of the n^{th} orbit?
- (a) $0.5a_0$ (b) $1.0a_0$ (c) $1.5a_0$ (d) $2.0a_0$
15. The energy of the n^{th} orbit of a hydrogen atom is -13.6 eV . What will be the radius of the n^{th} orbit?
- (a) $0.5a_0$ (b) $1.0a_0$ (c) $1.5a_0$ (d) $2.0a_0$
16. The energy of the n^{th} orbit of a hydrogen atom is -13.6 eV . What will be the radius of the n^{th} orbit?
- (a) $0.5a_0$ (b) $1.0a_0$ (c) $1.5a_0$ (d) $2.0a_0$
17. The energy of the n^{th} orbit of a hydrogen atom is -13.6 eV . What will be the radius of the n^{th} orbit?
- (a) $0.5a_0$ (b) $1.0a_0$ (c) $1.5a_0$ (d) $2.0a_0$
18. The energy of the n^{th} orbit of a hydrogen atom is -13.6 eV . What will be the radius of the n^{th} orbit?
- (a) $0.5a_0$ (b) $1.0a_0$ (c) $1.5a_0$ (d) $2.0a_0$
19. The energy of the n^{th} orbit of a hydrogen atom is -13.6 eV . What will be the radius of the n^{th} orbit?
- (a) $0.5a_0$ (b) $1.0a_0$ (c) $1.5a_0$ (d) $2.0a_0$
20. The energy of the n^{th} orbit of a hydrogen atom is -13.6 eV . What will be the radius of the n^{th} orbit?
- (a) $0.5a_0$ (b) $1.0a_0$ (c) $1.5a_0$ (d) $2.0a_0$

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Kathleen Armour



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