

Experiments 1: Determining Heat Capacity and

Using the Heat Capacity (see configuration below their way)

calculated difference of temperature after 5 minutes & concentration of the solution and
 mass of solution & solution

percentage uncertainty: determine that concentration that the solution is used all about their degree of knowledge
 of a given temperature

Procedure

Experiment 1: Heat Capacity

Part 1: Determination of Heat Capacity

Heat capacity is the ability to absorb heat (measured in J/K) without changing temperature. Heat capacity is the amount of heat that must be added to a substance to raise its temperature by 1 K.

Heat capacity is a property of a substance and is not affected by the amount of substance. Heat capacity is a property of a substance and is not affected by the amount of substance.

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Experiment 2: Heat Capacity

Heat capacity is the ability to absorb heat (measured in J/K) without changing temperature. Heat capacity is the amount of heat that must be added to a substance to raise its temperature by 1 K.

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Mass of solution (g)	Initial temp (°C)	Final temp (°C)	Heat capacity (J/K)	Heat capacity (J/K)	Heat capacity (J/K)
10.0	20.0	20.0	10.0	10.0	10.0
20.0	20.0	20.0	20.0	20.0	20.0
30.0	20.0	20.0	30.0	30.0	30.0
40.0	20.0	20.0	40.0	40.0	40.0

Figure 1



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S Ashworth



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