Name	Clans	Date	
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## Section: Concentration and Molarity

Complete each statement below by choosing a term from the following list. Use each term only once.

parts per million	moles	molarity	ratios	Liter
1	s the quantity o	f solute in a s	pecific quantity	y of solvent or
solution. One way o	of expressing c	oncentration i	s to use the un	it of measure
called	; it is the	ratio of	s	mount (in
moles) to	volur	ne (in liters),	or simply	
solute per L of solu	tion. Table 1 is	your textbo	ok lists other	concentration
units or	, such a	s	(m = n	nol solute/kg
solvent) and	q)	pm = mL solv	ate/1 000 000.0	mL solution).

## Solve the following problems and write your answer in the space provided.

2. What is the molarity of a hydrochloric acid solution that has a volume of 1500 mL and contains 441 g of HCl?

3. What is the molarity of a sugar solution that has a volume of 0.500 L and contains 17.1 g of ordinary sugar ( $C_{12}H_{22}O_{11}$ )?

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