

CHEMISTRY 101  
Honor Exam I  
February 9, 2017  
Dr. D. DeCoste

Name \_\_\_\_\_

Signature \_\_\_\_\_

T.A. \_\_\_\_\_

This exam contains 17 questions on 6 numbered pages. Check now to make sure you have a complete exam. You have one hour and thirty minutes to complete the exam. Determine the best answer to the first 15 questions and enter these on the special answer sheet. Also, circle your responses in this exam booklet. Show all of your work and provide complete answers to questions 16 and 17.

1-15	(20 pts.)	_____
16	(15 pts.)	_____
17	(15 pts.)	_____
Total	(60 pts.)	_____

Useful Information:

Always assume ideal behavior for gases (unless explicitly told otherwise).

$$PV = nRT$$

$$R = 0.08206 \text{ L}\cdot\text{atm}/\text{mol}\cdot\text{K}$$

$$\text{K} = ^\circ\text{C} + 273$$

$$\text{Avogadro's number} = 6.022 \times 10^{23}$$

# Chem 1midterm Exam Sample

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*Abstract Bulletin of the Institute of Paper Chemistry* ,1987

**Chemical News and Journal of Physical Science** ,1896

*American Druggists' Circular and Chemical Gazette* ,1889

Standard Methods of Chemical Analysis: The elements.-v.2.

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