

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____
(a) how many atoms of Fe are in one formula unit
(b) how many sulfate ions can be attached to the iron atom
(c) the charge on each Fe ion
(d) the total positive charge of the formula unit
2. C The result of changing a subscript in a correctly written chemical formula is to _____
(a) change the number of moles represented by the formula
(b) change the charges on the other ions in the compound
(c) change the formula so that it no longer represents the compound it previously represented
(d) have no effect on the formula
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
4 elements a. How many elements make up this compound?
6 oxygen atoms b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
21 atoms c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
 4.2×10^{24} atoms d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
4. How many atoms are present in each of these formula units?
11 atoms a. $Ca(HCO_3)_2$
45 atoms b. $C_{12}H_{22}O_{11}$
10 atoms c. $Fe(ClO_2)_3$
9 atoms d. $Fe(ClO_3)_2$
5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
 H_2SO_3 c. What is the formula for sulfurous acid?
phosphoric acid d. What is the name for the acid H_3PO_4 ?

Chemical Formulas And Chemical Compounds Chapter 7 Test

IM Harris



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Table of Contents Chemical Formulas And Chemical Compounds Chapter 7 Test

1. Understanding the eBook Chemical Formulas And Chemical Compounds Chapter 7 Test
 - The Rise of Digital Reading Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Formulas And Chemical Compounds Chapter 7 Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Personalized Recommendations
 - Chemical Formulas And Chemical Compounds Chapter 7 Test User Reviews and Ratings

- Chemical Formulas And Chemical Compounds Chapter 7 Test and Bestseller Lists
- 5. Accessing Chemical Formulas And Chemical Compounds Chapter 7 Test Free and Paid eBooks
 - Chemical Formulas And Chemical Compounds Chapter 7 Test Public Domain eBooks
 - Chemical Formulas And Chemical Compounds Chapter 7 Test eBook Subscription Services
 - Chemical Formulas And Chemical Compounds Chapter 7 Test Budget-Friendly Options
- 6. Navigating Chemical Formulas And Chemical Compounds Chapter 7 Test eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Formulas And Chemical Compounds Chapter 7 Test Compatibility with Devices
 - Chemical Formulas And Chemical Compounds Chapter 7 Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Highlighting and Note-Taking Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Interactive Elements Chemical Formulas And Chemical Compounds Chapter 7 Test
- 8. Staying Engaged with Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Formulas And Chemical Compounds Chapter 7 Test
- 9. Balancing eBooks and Physical Books Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Formulas And Chemical Compounds Chapter 7 Test
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Setting Reading Goals Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Fact-Checking eBook Content of Chemical Formulas And Chemical Compounds Chapter 7 Test
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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