| | UNIT 4: Chemical Reactions Note Taking Document |
|-----------|---|
| Gruid | elines for making and writing observations: |
| | Describe as many properties as you can. |
| | Record detailed information that might help a chemist identify the substance. |
| | Use descriptive words (such as thick, pink liquid) rather than imprecise language. |
| | For example, DO NOT SAY, "light-colored stuff." |
| 4 | Avoid opinions (such as looks gross) |
| Voc | abulary: |
| 35.00 | something that can be about a substance |
| | such as its color, smell, or boiling point. |
| t C | hemists rely on some properties that can only be observed under |
| - | such as and points. |
| 2. D | Fferent substances have melting and boiling points. Therefore, it |
| co. | a substance. |
| ъ | is the temperature at which a solid will begin to melt |
| -04 | nd turn into a liquid |
| 4 | is the temperature at which a liquid will begin to boil and |
| | |
| | ern into vapor. |
| tu | |
| +. PHY | SICAL CHANGE |
| + PHY | SICAL CHANGE Physical changes may change the state of matter but the substance |
| +. PHY | SICAL CHANGE Physical changes may change the state of matter but the substance change. |
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| PHY | SICAL CHANGE Physical changes may change the state of matter but the substance |
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| PHY L | SICAL CHANGE Physical changes may change the state of matter but the substance change Atoms do change No new form The state of matter or shape may Can be MICAL CHANGE Chemical changes create an entirely substance. Referred to as Brings about something The chemical bonds of the molecules and they reform in a different manner. |

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Dr. Abhishek M

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