

5

ELECTRONS IN ATOMS

Practice Problems

In your notebook, solve the following problems.

SECTION 5.1 MODELS OF THE ATOM

- How many sublevels are in the following principal energy levels?
 - $n = 1$
 - $n = 2$
 - $n = 3$
 - $n = 4$
 - $n = 5$
 - $n = 6$
- How many orbitals are in the following sublevels?
 - 1s sublevel
 - 5s sublevel
 - 4d sublevel
 - 4f sublevel
 - 7s sublevel
 - 3p sublevel
 - fifth principal energy level
 - 6d sublevel
- What are the types of sublevels and number of orbitals in the following energy levels?
 - $n = 1$
 - $n = 2$
 - $n = 3$
 - $n = 4$
 - $n = 5$

SECTION 5.2 ELECTRON ARRANGEMENT IN ATOMS

- Write a complete electron configuration of each atom.
 - hydrogen
 - vanadium
 - magnesium
 - barium
 - bromine
 - sulfur
 - krypton
 - arsenic
 - radon

SECTION 5.3 PHYSICS AND THE QUANTUM MECHANICAL MODEL

- What is the wavelength of the radiation whose frequency is $5.00 \times 10^{15} \text{ s}^{-1}$? In what region of the electromagnetic spectrum is this radiation?
- An inexpensive laser that is available to the public emits light that has a wavelength of 670 nm. What are the color and frequency of the radiation?
- What is the energy of a photon whose frequency is $2.22 \times 10^{14} \text{ s}^{-1}$?
- What is the frequency of a photon whose energy is $6.00 \times 10^{-19} \text{ J}$?
- Arrange the following types of electromagnetic radiation in order of increasing frequency.
 - infrared
 - cosmic rays
 - visible light
 - radio waves
 - microwaves
 - ultraviolet
- Suppose that your favorite AM radio station broadcasts at a frequency of 1500 kHz. What is the wavelength in meters of the radiation from the station?

Chapter 5 Electrons In Atoms Workbook Answers

**Yi-Noo Tang, Wendy L. Keeney-
Kennicutt**



Chapter 5 Electrons In Atoms Workbook Answers:

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Chapter 5 Electrons In Atoms Workbook Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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