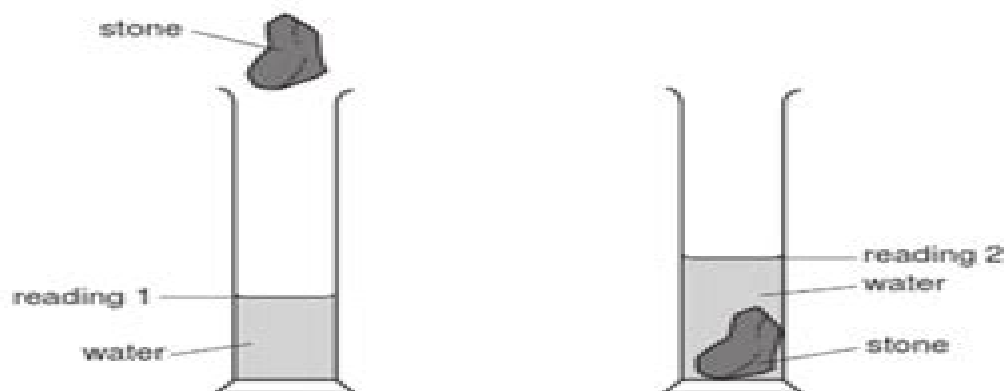


- 1 A student wishes to determine the density of an irregularly-shaped stone.

First he finds the mass of the stone. Next he lowers the stone into a measuring cylinder containing water.

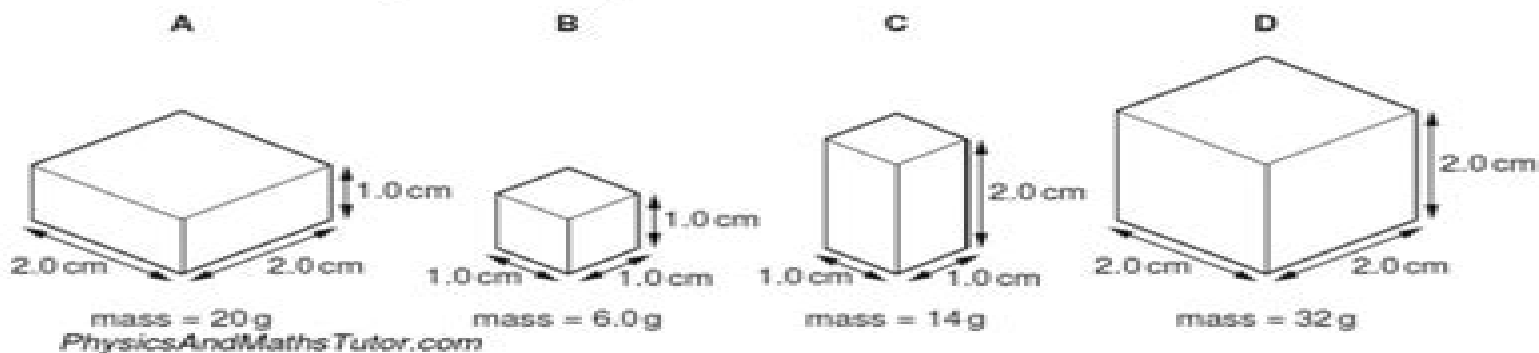
The diagrams show the measuring cylinder before and after the stone is lowered into it.



How should the student calculate the density of the stone?

- A mass of stone \times reading 2
B mass of stone \times (reading 2 – reading 1)
C mass of stone $+$ reading 2
D mass of stone $+$ (reading 2 – reading 1)
- 2 The diagrams show the dimensions and masses of four regular solid objects. The objects are made from different metals.

Which metal has the greatest density?



Density Multiple Choice Questions With Answers

Matthew Moncrieff Pattison Muir



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