

Name: _____

Double Replacement Lab

Reminder – Goggles must be worn at all times in the lab!

PRE-LAB DISCUSSION:

A PRECIPITATE is an insoluble substance formed by the reaction between chemicals in solution. A list of reactions will be given to you. The reactants are on the left side of the equation, and the products are on the right. The precipitate will be found on the product side. Keep in mind that an equation gives a condensed picture of what happens when two chemicals react.

PURPOSE:

The purpose of the lab is to observe semi-micro scale precipitation reactions, and to use a solubility table to identify the insoluble product, when there is one.

PROCEDURE:

Predict the products and balance the following double replacement reactions. Then, identify the precipitates using the table on page 2. **You may want to do your work step on a scratch piece of paper and attach to lab later.**

<u>Double Replacement Reactions</u>					Precipitates (if any)
A.	NaCl	+	AgNO ₃	→	
B.	KI	+	AgNO ₃	→	
C.	NaOH	+	AgNO ₃	→	
D.	Na ₃ PO ₄	+	AgNO ₃	→	
E.	NaCl	+	Pb(NO ₃) ₂	→	
F.	KI	+	Pb(NO ₃) ₂	→	
G.	NaOH	+	Pb(NO ₃) ₂	→	
H.	Na ₃ PO ₄	+	Pb(NO ₃) ₂	→	
I.	NaCl	+	CuSO ₄	→	
J.	KI	+	CuSO ₄	→	
K.	NaOH	+	CuSO ₄	→	
L.	Na ₃ PO ₄	+	CuSO ₄	→	

Double Replacement Reaction Lab Key

**Russo Steve Silver Michael, Steve
Russo**



Double Replacement Reaction Lab Key:

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forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance *Safety-Scale Lab Exp Biochem 2e* Spencer L. Seager, Michael R. Slabaugh, 1994-05 CHEMISTRY EXPERIMENTS James Signorelli, 2014-09-19 Gifted and talented students and any student interested in pursuing a science major in college needs a rigorous program to prepare them while they are still in high school This book utilizes a format where the application of several disciplines science math and language arts principles are mandated Each lab concludes with either an essay or a detailed analysis of what happened and why it happened This format is based on the expectations of joining a university program or becoming an industrial science professional The ideal student lab report would be written in a lab research notebook and then the essay or final analysis is done on a word processor to allow for repeat editing and corrections The research notebook has all graph pages a title section and a place for the students and their assistants to sign and witness that exercise The basic mechanics of the lab report title purpose procedure diagrams data table math and calculations observations and graphs are handwritten into the book The conclusion is done on a word processor MS Word which allows the instructor to guide the student in writing and editing a complete essay using the MLA format When the final copy is completed the essay is printed and inserted into the lab notebook for grading At the end of the term the student has all their labs in one place for future reference These lab notebooks can be obtained for as little as 3 00 per book This is money well spent In our district the Board of Education buys the books for each student The BOE sees these books as expendable but necessary materials for all science and engineering instruction Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Steps to Doing Science Kristin Tuttle Bump, 2008 **Inquiry-based Experiments in Chemistry** Valerie Ludwig Lechtanski, 2000 Inquiry Based Experiments in Chemistry is an alternative to those cookbook style lab manuals providing a more accurate and realistic experience of scientific investigation and thought for the high school chemistry or physical science student **Addison-Wesley Small-scale Chemistry** Dennis D. Staley, Edward L. Waterman, 1995 *Fundamentals of Chemistry in the Laboratory* Ralph A. Burns, 1999 **Novice Teacher Action** Anne Liu Kern, 2007 Carolina Science and Math Carolina Biological Supply Company, 2003 **Chemistry in the Laboratory** James M. Postma, Julian L. Roberts, J. Leland Hollenberg, 2009-08-07 For nearly 40 years Chemistry in the Laboratory has been meeting the needs of teachers and students This new edition builds on that legacy while addressing cutting edge trends in the chemistry laboratory including

forensic chemistry and environmental and green chemistry As always the new edition of Chemistry in the Laboratory offers precise easy to follow instructions helpful illustrations and an emphasis throughout on laboratory safety Again throughout a Consider This feature encourages users to expand the principles of the experiment into interesting applications open ended experiments or unexplored corners Most experiments in the manual can be completed in one lab session but some can be linked or extended for a multi lab project Prentice Hall Chemistry ,2000 Authentic Practice Chemistry David G. White,2000 *Introductory Chemistry* Steve Russo, Mike Silver,2002 Steve Russo and Mike Silver turn chemistry into a memorable story that engages readers and provides the context they need to understand and remember core concepts The book builds interesting applications and well designed illustrations into the narrative to get and hold attention then builds confidence with integrated active learning activities Readers make the connections between concepts and the problem solving techniques they need to master as they read The new edition strengthens this conceptual approach and presents additional quantitative techniques in key areas Readers will find enhanced support for quantitative problem solving and more challenging questions at the end of each chapter in addition to the wealth of technology based support on The Chemistry Place™ Special Edition and on The Chemistry of Life CD ROM For college instructors and students **A Concrete Stoichiometry Unit for High School Chemistry** Jennifer Louise Pakkala,2006 *Essentials of Introductory Chemistry* Russo Steve Silver Michael, Steve Russo,2001-12 Introductory Chemistry Third Edition helps readers master the quantitative skills and conceptual understanding they need to gain a deep understanding of chemistry Unlike other books on the market that emphasize rote memory of problem solving algorithms Introductory Chemistry takes a conceptual approach with the idea that focusing on the concepts behind chemical equations helps readers become more proficient problem solvers What Is Chemistry The Numerical Side of Chemistry The Evolution of Atomic Theory The Modern Model of the Atom 1 Chemical Bonding and Nomenclature The Shape of Molecules Chemical Reactions Stoichiometry and the Mole The Transfer of Electrons from One Atom to Another in a Chemical Reaction Intermolecular Forces and the Phases of Matter What If There Were No Intermolecular Forces The Ideal Gas Solutions When Reactants Turn into Products Chemical Equilibrium Electrolytes Acids and Bases For all readers interested in introductory chemistry **Prentice Hall Science Explorer** Michael J. Padilla,2002

The book delves into Double Replacement Reaction Lab Key. Double Replacement Reaction Lab Key is a vital topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Double Replacement Reaction Lab Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Double Replacement Reaction Lab Key
- Chapter 2: Essential Elements of Double Replacement Reaction Lab Key
- Chapter 3: Double Replacement Reaction Lab Key in Everyday Life
- Chapter 4: Double Replacement Reaction Lab Key in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Double Replacement Reaction Lab Key. The first chapter will explore what Double Replacement Reaction Lab Key is, why Double Replacement Reaction Lab Key is vital, and how to effectively learn about Double Replacement Reaction Lab Key.

3. In chapter 2, the author will delve into the foundational concepts of Double Replacement Reaction Lab Key. The second chapter will elucidate the essential principles that must be understood to grasp Double Replacement Reaction Lab Key in its entirety.

4. In chapter 3, the author will examine the practical applications of Double Replacement Reaction Lab Key in daily life. The third chapter will showcase real-world examples of how Double Replacement Reaction Lab Key can be effectively utilized in everyday scenarios.

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6. In chapter 5, the author will draw a conclusion about Double Replacement Reaction Lab Key. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Double Replacement Reaction Lab Key.

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