Name	Date	Period
:- 1		

#### Double-Replacement Precipitation Lab

<u>Introduction</u>-Double-Replacement reactions occur when two ionic solutions mix to form an insoluble solid. If all cations and anions are soluble then no reaction will occur. During the lab procedure, several ionic solutions will be mixed to determine if a precipitate reaction will occur. Those reaction that do not form a precipitate will be marked for No Reaction (NR). Students will practice writing and balancing double-replacement reactions.

#### Pre-Lab questions:

- How would you describe a double replacement reaction to a friend who is struggling with reaction types?
- 2. What is a precipitate?

#### Minterials:

- · 0.2M sodium beomide, NaBr
- 0.2M sodium carbonate, Na<sub>2</sub>CO<sub>3</sub>
- 0.2M sodium phosphate, Na<sub>3</sub>PO<sub>4</sub>
- 0.2M sodium sulfate, Na<sub>2</sub>SO<sub>4</sub>
- · 0.2M sodium hydroxide, NaOH
- Transparency paper

- · 0.2M lithium chloride, LiCl.
- 0.2M copper(II) chloride, CuCl<sub>2</sub>
- 0.2M zine chloride, ZnCl<sub>2</sub>
- · 0.2M strontium chloride, SrCl<sub>2</sub>
- 0.2M barium chloride BaCl<sub>2</sub>
- · Background paper Template

Safety: You MUST wear goggles during this lab activity

#### Procedure:

- Place the overhead sheet over the background paper.
- Using the Data Table as a guide, mix every possible pair of solutions in a separate square of the background paper. <u>Use I or 2 drops of each solution</u>. Do NOT contaminate the individual pipets with different solutions.
- 3. Mix the solutions with a rinsed toothpick if you are uncertain if a precipitate formed. Rinse the toothpick after EACH mixing be careful of which end of the rod you use so that you do not touch the chemicals or contaminate your samples!
- 4. Observe each mixture carefully for signs of a precipitate. (Some precipitates are light in color and hard to see so you may want to slide the transparency over the black lab top on occasion.) Record your observations (presence or absence of a precipitate, and other characteristics such as color, texture, etc.) in the Data Table. If nothing happens, write NR for no reaction.
- Rinse and wige the transparency sheet into the sink. Dry the sheet.

Observations: Write the observations you made from combining these materials in the spaces in the table below. Note any color changes or precipitate (cloudiness/solid). If no change is observed write NR for no reaction.

	NaBr	Na <sub>2</sub> CO <sub>3</sub>	NaoPO <sub>4</sub>	NagSO <sub>4</sub>	NaOH
LiCI					
CuClz					
ZnCl <sub>2</sub>					
SrCl <sub>2</sub>					
BaCl <sub>2</sub>					

# **Double Replacement Reaction Lab 27 Answers**

**Sebastian Brünink** 

#### **Double Replacement Reaction Lab 27 Answers:**

Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02
Regents Exams and Answers: Chemistry-Physical Setting Revised Edition Barron's Educational Series, Albert
Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the
Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This
book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough
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web which of the following statements about nervous systems is true a as nervous systems evolved there was a trend toward a decreasing number of interneurons b as nervous systems evolved the hindbrain and especially the cerebellum played an increasingly more dominant role c

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web the nervous system enables humans to react to their surroundings and to coordinate their behaviour it comprises millions of neurones and uses electrical impulses to communicate very quickly part of

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web unit 7 answer key unit 7 answer key draw and label a myelinated neuron showing the cell body dendrite axon axon terminal schwann cells and nodes of ranvier provide a brief description of the function of each labeled structure beside its label please be neat below is a diagram of a reflex arc

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