

Name: \_\_\_\_\_

## Double Replacement Lab

Reminder – Goggles must be worn at all times in the lab!

### PRE-LAB DISCUSSION:

A PRECIPITATE is an insoluble substance formed by the reaction between chemicals in solution. A list of reactions will be given to you. The reactants are on the left side of the equation, and the products are on the right. The precipitate will be found on the product side. Keep in mind that an equation gives a condensed picture of what happens when two chemicals react.

### PURPOSE:

The purpose of the lab is to observe semi-micro scale precipitation reactions, and to use a solubility table to identify the insoluble product, when there is one.

### PROCEDURE:

Predict the products and balance the following double replacement reactions. Then, identify the precipitates using the table on page 2. **You may want to do your work step on a scratch piece of paper and attach to lab later.**

<u>Double Replacement Reactions</u>					<b>Precipitates (if any)</b>
A.	NaCl	+	AgNO <sub>3</sub>	→	
B.	KI	+	AgNO <sub>3</sub>	→	
C.	NaOH	+	AgNO <sub>3</sub>	→	
D.	Na <sub>3</sub> PO <sub>4</sub>	+	AgNO <sub>3</sub>	→	
E.	NaCl	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
F.	KI	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
G.	NaOH	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
H.	Na <sub>3</sub> PO <sub>4</sub>	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
I.	NaCl	+	CuSO <sub>4</sub>	→	
J.	KI	+	CuSO <sub>4</sub>	→	
K.	NaOH	+	CuSO <sub>4</sub>	→	
L.	Na <sub>3</sub> PO <sub>4</sub>	+	CuSO <sub>4</sub>	→	

# Double Replacement Reaction Lab 2answers

**Terry C. Jones**



## **Double Replacement Reaction Lab 2answers:**

CHEMISTRY EXPERIMENTS James Signorelli, 2014-09-19 Gifted and talented students and any student interested in pursuing a science major in college needs a rigorous program to prepare them while they are still in high school This book utilizes a format where the application of several disciplines science math and language arts principles are mandated Each lab concludes with either an essay or a detailed analysis of what happened and why it happened This format is based on the expectations of joining a university program or becoming an industrial science professional The ideal student lab report would be written in a lab research notebook and then the essay or final analysis is done on a word processor to allow for repeat editing and corrections The research notebook has all graph pages a title section and a place for the students and their assistants to sign and witness that exercise The basic mechanics of the lab report title purpose procedure diagrams data table math and calculations observations and graphs are handwritten into the book The conclusion is done on a word processor MS Word which allows the instructor to guide the student in writing and editing a complete essay using the MLA format When the final copy is completed the essay is printed and inserted into the lab notebook for grading At the end of the term the student has all their labs in one place for future reference These lab notebooks can be obtained for as little as 3 00 per book This is money well spent In our district the Board of Education buys the books for each student The BOE sees these books as expendable but necessary materials for all science and engineering instruction E3 Chemistry Review Book - 2018 Home Edition (Answer Key Included) Effiong Eyo, 2017-10-20 With Answer Key to All Questions Chemistry students and homeschoolers Go beyond just passing Enhance your understanding of chemistry and get higher marks on homework quizzes tests and the regents exam with E3 Chemistry Review Book 2018 With E3 Chemistry Review Book students will get clean clear engaging exciting and easy to understand high school chemistry concepts with emphasis on New York State Regents Chemistry the Physical Setting Easy to read format to help students easily remember key and must know chemistry materials Several example problems with solutions to study and follow Several practice multiple choice and short answer questions at the end of each lesson to test understanding of the materials 12 topics of Regents question sets and 3 most recent Regents exams to practice and prep for any Regents Exam This is the Home Edition of the book Also available in School Edition ISBN 978 197836229 The Home Edition contains an answer key section Teachers who want to recommend our Review Book to their students should recommend the Home Edition Students and parents whose school is not using the Review Book as instructional material as well as homeschoolers should buy the Home Edition The School Edition does not have answer key in the book A separate answer key booklet is provided to teachers with a class order of the book Whether you are using the school or Home Edition our E3 Chemistry Review Book makes a great supplemental instructional and test prep resource that can be used from the beginning to the end of the school year PLEASE NOTE Although reading contents in both the school and home editions are identical there are slight differences in question numbers choices and pages between the two editions

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**Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today** Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

**Safety-Scale Lab Exp Biochem 2e** Spencer L. Seager, Michael R. Slabaugh, 1994-05

**Exploring General Chemistry in the Laboratory** Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science

*Regents Exams and Answers: Chemistry--Physical Setting Revised Edition* Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test

Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day

Whole-class Inquiry Dennis W. Smithenry,Joan Gallager-Bolos,2009 In response to requests from science education professionals this is the perfect vehicle for implementing and assessing this concept of whole class inquiry in your classroom This is a must have package for preservice and inservice middle and high school science teachers     **Chemistry** Eugene LeMay, Jr.,Herbert Beall,Karen M. Robblee,Douglas C. Brower,Prentice-Hall Staff,2002-02     *40 Low-Waste, Low-Risk Chemistry Labs* David Dougan,1997 Builds essential process and thinking skills Investigates central chemistry concepts Features procedures for purchase storage use and disposal of chemicals     *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science* ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction     Study Guide/Selected Solutions Manual Julie R. Frentrup,2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book     *Prentice Hall Chemistry* ,2000     Prentice Hall Science Explorer Michael J. Padilla,2002     Improving Student Comprehension in Chemistry Laboratories Tracy Lynn Haroff,2006     *Authentic Practice Chemistry* David G. White,2000

**General Catalog** Iowa State University,1917     **Inquiry-based Experiments in Chemistry** Valerie Ludwig Lechtanski,2000 Inquiry Based Experiments in Chemistry is an alternative to those cookbook style lab manuals providing a more accurate and realistic experience of scientific investigation and thought for the high school chemistry or physical science student     **Journal of the Society of Chemical Industry** Society of Chemical Industry (Great Britain),1915 Includes list of members 1882 1902 and proceedings of the annual meetings and various supplements     *An Introductory Guide to EC Competition Law and Practice* Valentine Korah,1994

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