

Name: \_\_\_\_\_

## Double Replacement Lab

Reminder – Goggles must be worn at all times in the lab!

### PRE-LAB DISCUSSION:

A PRECIPITATE is an insoluble substance formed by the reaction between chemicals in solution. A list of reactions will be given to you. The reactants are on the left side of the equation, and the products are on the right. The precipitate will be found on the product side. Keep in mind that an equation gives a condensed picture of what happens when two chemicals react.

### PURPOSE:

The purpose of the lab is to observe semi-micro scale precipitation reactions, and to use a solubility table to identify the insoluble product, when there is one.

### PROCEDURE:

Predict the products and balance the following double replacement reactions. Then, identify the precipitates using the table on page 2. **You may want to do your work step on a scratch piece of paper and attach to lab later.**

<u>Double Replacement Reactions</u>					Precipitates (if any)
A.	NaCl	+	AgNO <sub>3</sub>	→	
B.	KI	+	AgNO <sub>3</sub>	→	
C.	NaOH	+	AgNO <sub>3</sub>	→	
D.	Na <sub>3</sub> PO <sub>4</sub>	+	AgNO <sub>3</sub>	→	
E.	NaCl	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
F.	KI	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
G.	NaOH	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
H.	Na <sub>3</sub> PO <sub>4</sub>	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
I.	NaCl	+	CuSO <sub>4</sub>	→	
J.	KI	+	CuSO <sub>4</sub>	→	
K.	NaOH	+	CuSO <sub>4</sub>	→	
L.	Na <sub>3</sub> PO <sub>4</sub>	+	CuSO <sub>4</sub>	→	

# Double Replacement Reaction Lab Answers 10 2

**Albert A Gayle**



## Double Replacement Reaction Lab Answers 10 2:

Technical Abstract Bulletin Defense Documentation Center (U.S.),1961-07      **Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today** Seager/Slabaugh,2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application      **Basic Concepts of Chemistry** Leo J. Malone,Theodore Dolter,2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance      **Chemistry** McGraw-Hill Staff,2001-07      **Safety-Scale Lab Exp Biochem 2e** Spencer L. Seager,Michael R. Slabaugh,1994-05      **Nuclear Science Abstracts** ,1967      **Inquiry-based Experiments in Chemistry** Valerie Ludwig Lechtanski,2000 Inquiry Based Experiments in Chemistry is an alternative to those cookbook style lab manuals providing a more accurate and realistic experience of scientific investigation and thought for the high school chemistry or physical science student      **U.S. Government Research & Development Reports** ,1970      **Chemical Abstracts** ,1925  
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## **Table of Contents Double Replacement Reaction Lab Answers 10 2**

1. Understanding the eBook Double Replacement Reaction Lab Answers 10 2
  - The Rise of Digital Reading Double Replacement Reaction Lab Answers 10 2
  - Advantages of eBooks Over Traditional Books
2. Identifying Double Replacement Reaction Lab Answers 10 2
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Double Replacement Reaction Lab Answers 10 2
  - User-Friendly Interface
4. Exploring eBook Recommendations from Double Replacement Reaction Lab Answers 10 2
  - Personalized Recommendations
  - Double Replacement Reaction Lab Answers 10 2 User Reviews and Ratings
  - Double Replacement Reaction Lab Answers 10 2 and Bestseller Lists
5. Accessing Double Replacement Reaction Lab Answers 10 2 Free and Paid eBooks

- Double Replacement Reaction Lab Answers 10 2 Public Domain eBooks
  - Double Replacement Reaction Lab Answers 10 2 eBook Subscription Services
  - Double Replacement Reaction Lab Answers 10 2 Budget-Friendly Options
6. Navigating Double Replacement Reaction Lab Answers 10 2 eBook Formats
- ePub, PDF, MOBI, and More
  - Double Replacement Reaction Lab Answers 10 2 Compatibility with Devices
  - Double Replacement Reaction Lab Answers 10 2 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Double Replacement Reaction Lab Answers 10 2
  - Highlighting and Note-Taking Double Replacement Reaction Lab Answers 10 2
  - Interactive Elements Double Replacement Reaction Lab Answers 10 2
8. Staying Engaged with Double Replacement Reaction Lab Answers 10 2
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Double Replacement Reaction Lab Answers 10 2
9. Balancing eBooks and Physical Books Double Replacement Reaction Lab Answers 10 2
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Double Replacement Reaction Lab Answers 10 2
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Double Replacement Reaction Lab Answers 10 2
- Setting Reading Goals Double Replacement Reaction Lab Answers 10 2
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Double Replacement Reaction Lab Answers 10 2
- Fact-Checking eBook Content of Double Replacement Reaction Lab Answers 10 2
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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