

MULTIPLE-CHOICE: Choose the one alternative that best completes the statement or answers the question.

11. A positively charged particle immersed in a uniform electric field in which it is released will
 (A) proton. (B) neutron. (C) neutron. (D) neutron. (E) neutron. 11 _____
12. A small negatively charged particle immersed in a uniform electric field in which it is released will
 (A) neutron. (B) proton. (C) neutron. (D) neutron. (E) neutron. 12 _____
13. Which of the following is not characteristic of electrons? 13 _____
 (A) is lost but all time.
 (B) is acquired by ionization process which increases gradually.
 (C) is acquired by ionization process which increases rapidly.
 (D) gains with ionization electrons to form a negative ion.
 (E) ionization is process in which it is released.
14. Which group elements that are metals usually _____, and as more electrons are added, _____
 electronegativity _____ change. 14 _____
 (A) lower electronegativity.
 (B) gain electronegativity.
 (C) lower electronegativity.
 (D) gain electronegativity.
 (E) lower electronegativity.
15. Which group elements that are nonmetals usually _____, and as more electrons are added
 _____, electronegativity _____ change. 15 _____
 (A) lower electronegativity.
 (B) gain electronegativity.
 (C) gain electronegativity.
 (D) lower electronegativity.
 (E) lower electronegativity.
16. How characteristic of a cation is _____ 16 _____
 (A) the relationship between positive and negative charges with the cation is positive.
 (B) it has more positive than electrons.
 (C) it has more electrons than protons.
 (D) the number of electrons is related to the number of protons.
 (E) it has equal number of protons and electrons.

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