

Quiz & Worksheet - Double Displacement Reaction

<http://study.com/academy/practice/quiz-worksheet-double-displacement-reaction.html>

1. What type of reaction is the following?



☐ combustion

☐ gas formation

☐ precipitation

☐ neutralization

2. What are the ions and the correct charges for the following compound:



☐ Pb^{2+} Cl^{2-}

☐ Pb^{2-} Cl^{2-}

☐ Pb^{+} Cl^{2-}

☐ Pb^{2+} Cl^{-}

3. What are the correct products for the following reaction? (also include the correct coefficient for a balanced reaction)



☐ $\text{Ag}_2\text{SO}_4 + \text{Na}_2\text{NO}_3$

☐ $\text{Ag}_2\text{SO}_4 + \text{NaNO}_3$

☐ $\text{Ag}_2\text{SO}_4 + 2\text{NaNO}_3$

☐ $\text{AgSO}_4 + \text{NaNO}_3$

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Double Replacement Reactions Lab Answer Key

**Spencer L. Seager, Michael R.
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school chemistry or physical science student **Novice Teacher Action** Anne Liu Kern,2007 **Authentic Practice Chemistry** David G. White,2000 **Introductory Chemistry** Steve Russo, Mike Silver,2002 Steve Russo and Mike Silver turn chemistry into a memorable story that engages readers and provides the context they need to understand and remember core concepts The book builds interesting applications and well designed illustrations into the narrative to get and hold attention then builds confidence with integrated active learning activities Readers make the connections between concepts and the problem solving techniques they need to master as they read The new edition strengthens this conceptual approach and presents additional quantitative techniques in key areas Readers will find enhanced support for quantitative problem solving and more challenging questions at the end of each chapter in addition to the wealth of technology based support on The Chemistry Place tm Special Edition and on The Chemistry of Life CD ROM For college instructors and students

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