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Carpenter**



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web feb 19 2022 in a balanced chemical equation both the numbers of each type of atom and the total charge are the same on both sides equations ref 3 1 1 and ref 3 1 2 are balanced chemical equations what is different on each side of the equation is how the atoms are arranged to make molecules or ions

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web balancing equations the chemical equation described in section 4 1 is balanced meaning that equal numbers of atoms for each element involved in the reaction are represented on the reactant and product sides this is a requirement the equation must satisfy to be consistent with the law of conservation of matter

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web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide CaO s and H_2O l to form aqueous calcium hydroxide b is it possible to balance the equation if you incorrectly identify the product as CaOH aq and if so what is the equation

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web dec 2 2020 a balanced chemical equation tells you the amounts of reactants and products needed to satisfy the law of conservation of mass basically this means there are the same numbers of each type of atoms on the left side of the equation as there are on the right side of the equation

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web balance the following chemical equation $\text{Mg} + \text{OH}^- + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$ note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem do 4 problems

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