

Name: _____

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Gas Laws Test Review

Select the correct answer for the following questions. For any calculation you **MUST** show ALL work with units. For any multiple choice question you **MUST** list the textbook page number or section that help you select the correct answer.

- If temperature and pressure are held constant, the volume and number of moles of a gas are:
a. directly proportional b. independent of each other
c. not enough information given d. equal e. inversely proportional
- If volume and number of moles of a gas are held constant, the temperature and pressure are:
a. directly proportional b. independent of each other
c. not enough information given d. equal e. inversely proportional
- If temperature and number of moles of a gas are held constant, the volume and pressure are:
a. directly proportional b. independent of each other
c. not enough information given d. equal e. inversely proportional
- If pressure and number of moles of a gas are held constant, the volume and temperature are:
a. directly proportional b. independent of each other
c. not enough information given d. equal e. inversely proportional
- One mole of NH_3 at STP will have what volume?
a. 22.4 L b. 44g c. 1.0 L d. 24.5 L
- Which of the following statements is true about the kinetic molecular theory?
a) Pressure is due to the collisions of the gas particles with the walls of the container.
b) Adding an ideal gas to a closed container will cause an increase in temperature.
c) The volume of a gas particle is considered to be small about 0.1 mL.
d) Gas particles have random motion.
e) At least two of these statements are true.
- Which of the following statements is true concerning ideal gases?
a) The gas particles in a sample exert attraction for one another.
b) At STP 1.0L of Ar gas contains about twice the number of atoms as 1.0L of Ne gas.
c) A gas exerts pressure as a result of the collisions of the gas molecules with the walls of the container.
d) The temperature of the gas sample is indirectly related to the average velocity of the gas particles.
- Use the kinetic molecular theory of gases to predict what would happen to a closed sample of a gas whose temperature increased while its volume decreased.
a) Its pressure would decrease c) Its pressure would hold constant.
b) Its pressure would increase. d) the number of moles would increase.
- Perform the following conversions of pressure units: $5.0 \times 10^5 \text{ Pa} = \underline{\hspace{2cm}} \text{ atm}$
a. 4.9×10^4 b. 1.7×10^5 c. 4.3×10^4 d. 2.5×10^4 e. 9.8×10^4
- The pressure of a given mass of gas varies directly with the temperature when the volume remains constant.
a. Boyle's Law b. combined gas law c. Gay-Lussac's Law
d. ideal gas law e. Charles's Law
- If the temperature is held constant, what will happen to the pressure exerted by a gas if the volume of the container is decreased?
a. increase b. decrease c. remain the same d. increase or decrease (it depends)
- If the temperature of a sample of gas is decreased at constant pressure, the volume of the gas will:
a. increase b. decrease c. remain the same d. you cannot predict this.

Concept Review Section The Gas Laws Answer

**Michael J. Padilla, Ioannis
Miaoulis, Martha Cyr**



Concept Review Section The Gas Laws Answer:

Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change **Master the GED 2012** Peterson's, 2011-12-01 Master the GED is a comprehensive guide that offers the essential test prep and review material for the high school equivalency diploma test including practice tests subject review and expert tips on how to score high on each GED test Readers will find the GED information they need to know scoring and passing requirements how to prepare and what to expect on test day The Cosmos Jay M. Pasachoff, Alex Filippenko, 2019-07-11 Explains the fundamentals of astronomy together with the hottest current topics in this field such as exoplanets and gravitational waves **Master the GED 2013** Peterson's, 2012-06-29 Peterson's Master the GED 2013 offers thorough test preparation for individuals who have decided to get their high school diplomas This self tutor will show you what to expect while giving you the most effective practice with subjects you can presume to see on the actual exam The first part of the eBook covers essential information about the structure of the exam scoring and passing requirements scheduling and testing procedures and what you need to get ready for the exam Next comes a diagnostic practice test to help you see where your strengths and weaknesses are The following five parts offer thorough reviews of the subject matter for each test area of the GED Language Arts Writing Parts I and II Social Studies Science Language Arts Reading and Mathematics Each review includes test taking strategies and practice questions with detailed answer explanations The final part of this eBook comprises two full length tests so that you can apply your knowledge and newly learned strategies and practice for the actual GED Finally a word list in the appendix offers you a great tool to boost your vocabulary Physics John D. Cutnell, Kenneth W. Johnson, 2009-09-08 Designed for medical professionals who may struggle with making the leap to conceptual understanding and applying physics the eighth edition continues to build transferable problem solving skills It includes a set of features such as Analyzing Multiple Concept Problems Check Your Understanding Concepts Calculations and Concepts at a Glance This helps the reader to first identify the physics concepts then associate the appropriate mathematical equations and finally to work out an algebraic solution General Chemistry Key Concise Review Guide Sixth Edition Darrell D. Ebbing, 1999 CHEMISTRY SILBERBERG, 2003 **Holt Chemistry** Ralph Thomas Myers, 2004 *Focus on physical science* Charles H. Heimler, 1989 Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 Chemistry John S. Phillips, Cheryl Wistrom, 2000 **Evaluation Package for Cutnell and Johnson Physics 8E** Cutnell, 2008-10-28 **Prentice Hall Chemistry**, 2000 *Science Interactions 1* McGraw-Hill Education, 1998 **Railway Review**, 1923 *Harcourt Science: Physical science, [grade] 5, Units E and F, teacher's ed*, 2000 The Electrical Review, 1929 **Telegraphic Journal and Electrical Review**, 1929 *Science Explorer: Sound and Light* Michael J. Padilla, Ioannis Miaoulis, Martha Cyr, 2009 **Telegraphic Journal and Monthly Illustrated Review of Electrical Science**, 1924

The Top Books of the Year Concept Review Section The Gas Laws Answer The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels captivating the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the captivating narratives that have charmed audiences this year. Concept Review Section The Gas Laws Answer : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Concept Review Section The Gas Laws Answer : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens' "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and thrilling novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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