

Name \_\_\_\_\_

Class \_\_\_\_\_

Date \_\_\_\_\_

## Skills Worksheet

**Concept Review****Section: Concentration and Molarity**

Complete each statement below by choosing a term from the following list.  
Use each term only once.

concentration

moles

solute

solution

liter

parts per million

molality

molarity

ratios

1. \_\_\_\_\_ is the quantity of solute in a specific quantity of solvent or solution. One way of expressing concentration is to use the unit of measure called \_\_\_\_\_; it is the ratio of \_\_\_\_\_ amount (in moles) to \_\_\_\_\_ volume (in liters), or simply \_\_\_\_\_ of solute per L of solution. Table 1 in your textbook lists other concentration units or \_\_\_\_\_, such as \_\_\_\_\_ ( $m = \text{mol solute/kg solvent}$ ) and \_\_\_\_\_ ( $\text{ppm} = \text{mL solute/1 000 000.0 mL solution}$ ).

Solve the following problems and write your answer in the space provided.

2. What is the molarity of a hydrochloric acid solution that has a volume of 1500 mL and contains 441 g of HCl?
3. What is the molarity of a sugar solution that has a volume of 0.500 L and contains 17.1 g of ordinary sugar ( $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ )?

# Concept Review Section Solubility And Concentration

**JR Anderson**



## Concept Review Section Solubility And Concentration:

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examples and a Maths Toolkit to help students get to grips with the essential mathematical element of chemistry Frequent cross references highlight the connections between each strand of chemistry and explain the relationship between the topics so students can develop an understanding of the subject as a whole Digital formats and resources Chemistry3 is available for students and institutions to purchase in a variety of formats and is supported by online resources The e book offers a mobile experience and convenient access along with functionality tools navigation features and links that offer extra learning support [www.oxfordtextbooks.co.uk/ebooks](http://www.oxfordtextbooks.co.uk/ebooks) The e book also features interactive animations of molecular structures screencasts in which authors talk step by step through selected examples and key reaction mechanisms and self assessment activities for each chapter The accompanying online resources will also include for students DT Chapter 1 as an open access PDF DT Chapter summaries and key equations to download to support revision DT Worked solutions to the questions in the book The following online resources are also provided for lecturers DT Test bank of ready made assessments for each chapter with which to test your students DT Problem solving workshop activities for each chapter for you to use in class DT Case studies showing how instructors are successfully using Chemistry3 in digital learning environments and to support innovative teaching practices DT Figures and tables from the book

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*Future Fat Alternatives* Zong Meng, 2025-11-28 Future Fat Alternatives Concepts Structuring and Food Applications provides a deep dive into the strategies for the development of fat alternatives for food applications covering all structuring strategies including oleogels emulsion gels and foamed gels as well as cell cultured fats In addition this book details other applications such as additive manufacturing and fat alternatives in different food applications and discusses their nutritional benefits Divided into eight sections this book begins with an overview of the structural properties relationship in traditional fats then segues into the benefits and challenges of fat alternatives the

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Chemical Interactions ,2005 Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson Norman E. Griswold, Henry Fuller Holtzclaw,1988 Practical Druggist and Pharmaceutical Review of Reviews ,1904 *Practical Druggist and Pharmaceutical Review of Reviews* Benjamin Lillard,1904 **New York Review**

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