

Name: _____

Solve the unit conversion problem by cross cancelling units.

9.75 liters =
as milliliters

$$\frac{9.75 \text{ l}}{1} \times \frac{1000 \text{ ml}}{1 \text{ l}} = 9750 \text{ ml}$$

9.25 liters =
as milliliters

$$\frac{9.25 \text{ l}}{1} \times \frac{1000 \text{ ml}}{1 \text{ l}} = 9250 \text{ ml}$$

2.75 liters =
as centiliters

$$\frac{2.75 \text{ l}}{1} \times \frac{100 \text{ cl}}{1 \text{ l}} = 275 \text{ cl}$$

40 liters =
as milliliters

$$\frac{40 \text{ l}}{1} \times \frac{1000 \text{ ml}}{1 \text{ l}} = 40000 \text{ ml}$$

9.5 liters =
as centiliters

$$\frac{9.5 \text{ l}}{1} \times \frac{100 \text{ cl}}{1 \text{ l}} = 950 \text{ cl}$$

9.75 liters =
as centiliters

$$\frac{9.75 \text{ l}}{1} \times \frac{100 \text{ cl}}{1 \text{ l}} = 975 \text{ cl}$$

4.75 liters =
as milliliters

$$\frac{4.75 \text{ l}}{1} \times \frac{1000 \text{ ml}}{1 \text{ l}} = 4750 \text{ ml}$$

3.5 liters =
as milliliters

$$\frac{3.5 \text{ l}}{1} \times \frac{1000 \text{ ml}}{1 \text{ l}} = 3500 \text{ ml}$$

9.75 liters =
as milliliters

$$\frac{9.75 \text{ l}}{1} \times \frac{1000 \text{ ml}}{1 \text{ l}} = 9750 \text{ ml}$$

6.75 liters =
as milliliters

$$\frac{6.75 \text{ l}}{1} \times \frac{1000 \text{ ml}}{1 \text{ l}} = 6750 \text{ ml}$$

Dimensional Analysis Unit Conversion Answer Key

Neil D. Jespersen, Alison Hyslop



Dimensional Analysis Unit Conversion Answer Key:

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Malone, Theodore O. Dolter, 2012-04-13 Basic Concepts of Chemistry was originally written over thirty years ago to address the needs of general chemistry students with little or no background in chemistry Over time the text has evolved beyond purposes solely aimed at a preparatory chemistry course For some preparatory chemistry students a main sequence in general chemistry may follow but for others a semester of organic and biochemistry may follow Other students enroll to simply satisfy a basic science or chemistry requirement The text was written with a level and functionality designed to accommodate the needs of each of these varied groups of students Basic Concepts of Chemistry was designed with a flexibility that allows instructors to emphasize or omit certain clearly delineated sections The mission of the text has evolved in response to the increased diversity of students and the emphasis on outcomes assessment **Quality Labs for Small**

Brewers Merritt Waldron, 2020-08-03 Quality assurance and quality control QA QC is both a system and a state of mind In Quality Labs for Small Brewers author Merritt Waldron walks you step by step through the process of establishing and writing a quality program for your brewery Your quality policy should align with your company values and inculcate a quality first culture throughout your brewery Building an effective quality program will empower staff to directly influence the consistent production of safe quality beer from grain to glass A good quality program has many moving parts but it is

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