Math Skills, continued

- 2. Write down the specific heat equation. energy = $cm\Delta T$
- 3. Substitute the specific heat, mass, and temperature change values, and solve.

energy =
$$\left(\frac{4,186 \text{ J}}{\text{kg} \cdot \text{K}}\right) \times (200 \text{ kg}) \times (12 \text{ K})$$

energy =
$$10,000,000 \text{ J} = 1.0 \times 10^4 \text{ kJ}$$







Concept Review Using Heat Math Skills Continued

Huangqi Zhang

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