Concentration and Mol:	arity PhET Labs
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Hame

is added and when

Procedure: https://ghet.colorado.edu/sims/html/concentration/latest/concentration en.html

Part 1: Dissolution and Saturation

Take some time to play and familiarize yourself with the simulation. Click on everything. Move all



Concentration

evaporation occurs

How does the concentration change as solid solute is added?

2. How does the concentration change as additional water is added? concentration decreases

0.179

Comments without

3. How does the concentration change as evaporation occurs? concentration in creased

4. How do you know when a solution is soluroted? | concentration does not change (it also shows on screen)

6. When a solution is saturated, and additional solid solute is added, what happens? Solid does not dissolve and settles at bottom

7. Why do you think this is? solvent cannot hold any more solute

the sliders. Notice what happens to the concentration as solid solute

8. How does adding this additional solute change the concentration of this saturated solution? decreases concentration

9. How does evaporation change the concentration of a saturated solution? #0 change solution is already saturated

Part 2: Concentrated Solutions (same website as part 1 kmt click on the Solution option)

It Solution

10. Adding a concentrated solution ... describe a way to determine the concentration of the solution in the spigot. Write your plan heres _empty the container and add only one solution from the dropper and then use the

sensor to measure concentration

Delining .

Count distings Calculations for Pulsanium districts dis-

Course where

Parameter president per

11. Using your plan...how might you get that concentrated solution to become saturated? You can evaporate

12. Does your plan work for all the other solutions too? Yes Why? / Why Not? evaporation reduces the

amount of water (or solute) so concentration increases

Part 3: Molarity https://phet.colorada.edu/sims/html/molarity/latest/molarity_en.html

Mollaray (M) = dimount of solute (mol) solute of solution (L) Molarity is moles per Liter, that is, how many moles of solute (entire salt) is dissolved per Liter of solution.

First, determine the saturation concentration of each of the solutions, that is, how concentrated can you get each solution before the solution is saturated. If you can't determine the concentration using the simulation "Molarity", try using the simulation "Concentration" (You will use this information again in Port S. If your instructor requires it)



Motority

Cobalt (II) nitrate	5.640 M	Potassium chromate	3.350 M
Cobalt chloride	4.330 M	Nickel (II) chloride	5.210 M
Potassium dichromate	0.510 M	Copper sulfate	1.380 M
Gold (III) chloride	2.30 M	Potassium permanganate	0.480 IV

Concentration And Molarity Post Lab Exercises Answers

Hunt

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Laboratory Manual for Biotechnology and Laboratory Science Lisa A. Seidman, Mary Ellen Kraus, Diana Lietzke Brandner, Jeanette Mowery, 2022-12-23 Provides the basic laboratory skills and knowledge to pursue a career in biotechnology Written by four biotechnology instructors with over 20 years of teaching experience it incorporates instruction exercises and laboratory activities that the authors have been using and perfecting for years These exercises and activities help students understand the fundamentals of working in a biotechnology laboratory Building skills through an organized and systematic presentation of materials procedures and tasks the manual explores overarching themes that relate to all biotechnology workplaces including forensic clinical quality control environmental and other testing laboratories Features Provides clear instructions and step by step exercises to make learning the material easier for students There are Lab Notes for Instructors in the Support Material see tab below Emphasizes fundamental laboratory skills that prepare students for the industry Builds students skills through an organized and systematic presentation of materials procedures and tasks Updates reflect recent innovations and regulatory requirements to ensure students stay up to date Supplies skills suitable for careers in forensic clinical quality control environmental and other testing laboratories **Laboratory Manual for Biotechnology** Verma, Ashish S./ Das Surajit & Singh Anchal, Laboratory Manual in Biotechnology Students Methods in Biotechnology Seung-Beom Hong, M. Bazlur Rashid, Lory Z. Santiago-Vázguez, 2016-08-01 As rapid advances in biotechnology occur there is a need for a pedagogical tool to aid current students and laboratory professionals in biotechnological methods Methods in Biotechnology is an invaluable resource for those students and professionals Methods in Biotechnology engages the reader by implementing an active learning approach provided advanced study guestions as well as pre and post lab questions for each lab protocol These self directed study sections encourage the reader to not just perform experiments but to engage with the material on a higher level utilizing critical thinking and troubleshooting skills This text is broken into three sections based on level Methods in Biotechnology Advanced Methods in Biotechnology I and Advanced Methods in Biotechnology II Each section contains 14 22 lab exercises with instructor notes in appendices as well as an answer guide as a part of the book companion site This text will be an excellent resource for both students and laboratory professionals in the biotechnology field Foundation Course for NEET (Part 2): Chemistry Class 9 Lakhmir Singh & Manjit Kaur, Our NEET Foundation series is sharply focused for the NEET aspirants Most of the students make a career choice in the middle school and therefore choose their stream informally in secondary and formally in senior secondary schooling accordingly If you have decided to make a career in the medical profession you need not look any further Adopt this series for Class 9 and 10 today <u>Essential Methods of Instrumental Analysis</u> Frank M. Dunnivant, Jake W. Ginsbach, 2024-08-05 Intuitively organized textbook aligned to common analytical instrumentation courses for undergraduate students Through an analytical approach Essential Methods of Instrumental Analysis provides an expansive

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