

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

| | | | |
|------------|----------|---------|----------|
| immiscible | liquid | soluble | solution |
| insoluble | miscible | solvent | solvent |

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen in air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air's (4) soluble substance. A substance that does not dissolve is (5) insoluble. (6) Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about hydration in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is a solution in which the solvent is water. What is solvation?

Solvation is the process of surrounding solute particles with solvent particles to form a solution.

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Chemistry: Matter and Change • Chapter 15

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CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True _____ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Content Mastery Study Guide Answers

Susan K. Grove, Daisha J. CIPHER

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the authors' names.

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Statistics for Nursing Research - E-Book Susan K. Grove, Daisha J. CIPHER, 2024-01-20 Get the review and practice you need to critically appraise the statistical methods used in published research Statistics for Nursing Research 4th Edition helps you understand and interpret statistical methods sampling and measurement techniques and statistical analysis techniques This unique and practical workbook is divided into two parts to differentiate between basic and advanced statistical methods with dozens of hands on exercises that will help you practice working with the statistical methods needed for building an evidence based practice This workbook is an indispensable resource for any nursing student or practicing nurse wishing to gain competency in conducting statistical analyses Comprehensive coverage of basic statistical methods includes extensive practice with the most common techniques of sampling measurement and statistical analysis seen in nursing and health sciences literature Literature based approach uses key excerpts from published studies to reinforce learning through practical application 38 sampling measurement and statistical analysis exercises provide a practical review of both basic and advanced statistical techniques Study Questions in each chapter help you examine a statistical technique and apply concepts to an actual literature appraisal Online submission of answers for Questions for Further Application allows you to complete exercises in a web based format and then submit completed exercises to your instructor via email for grading Consistent format for all chapters enhances learning and enables quick review NEW More than half of the research

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