

Chemistry – Unit 7 Worksheet B

Write balanced chemical equations for the following reactions.

Hydrogen gas (collected in air) is also collected in (overhead) 1.0 minute (but immediately) over a 1.0

1. Acetylene (C_2H_2) reacts with oxygen (hydrogen) + hydrocarbon for three acetylene, a hydrocarbon:

$$2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$$

2. Calcium carbonate is decomposed (superheating) for three calcium carbonate and carbon dioxide:

$$\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$$

3. Barium sulfate reacts with water (for three barium hydroxide):

$$\text{BaSO}_4 + \text{H}_2\text{O} \rightarrow \text{Ba(OH)}_2$$

4. Methanol (CH_3OH) decomposes to form methane (CH_4) and carbon monoxide:

$$\text{CH}_3\text{OH} \rightarrow \text{CH}_4 + \text{CO}$$

5. Ethanol reacts with oxygen (O_2) to form ethane (ethanol) and carbon dioxide:

$$2\text{C}_2\text{H}_5\text{OH} + 3\text{O}_2 \rightarrow 2\text{C}_2\text{H}_6 + 3\text{H}_2\text{O}$$

6. Calcium sulfate decomposes (superheating) for three calcium sulfate and carbon dioxide:

$$\text{CaSO}_4 \rightarrow \text{CaO} + \text{SO}_2$$

7. Iron reacts with sulfur gas (Fe_2S_3) to form iron (III) sulfate and hydrogen gas:

$$2\text{Fe} + 3\text{H}_2\text{S}_3 \rightarrow \text{Fe}_2\text{S}_3 + 3\text{H}_2$$

8. A nitrogen-containing carbon compound, $\text{C}_2\text{H}_5\text{N}_2$, decomposes to form ethane, C_2H_6 , and nitrogen gas:

$$\text{C}_2\text{H}_5\text{N}_2 \rightarrow \text{C}_2\text{H}_6 + \text{N}_2$$

9. Phosphorus, P_4O_{10} , is formed when carbon monoxide reacts with phosphorus gas:

$$4\text{CO} + \text{P}_4 \rightarrow \text{P}_4\text{O}_{10} + 4\text{C}$$

Chemistry Unit 7 Ws3 Answers

Annelies Wilder-Smith



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