



Stoichiometry Balloon Races

Limiting and Excess Reactants

Introduction

Most stoichiometric) mode nation of reactions are performed using exact (stoichiometric) mode nation of reactions and products. In real life, however, many consensated processes for preparing companions are corried out using an excess amount of one treation (and thus a limiting amount of another reaction). This consensuals in assess the well-known reaction of sociams bicarbonate and acetic acid to illustrate the concepts of finiting and excess reactions. By comparing the amount of called to include persented when verying amounts of socials bicarbonate reactions in given amounts of acetic acid, stademant will be able to identify increasing that the finiting and excess reactions reall be able to identify

Concepts

- Suitchioreerry
- I limit time resultant.
- "Mode master."
- Excess reactions.

Materials

Acroic and, CH₂COOH, 1 M, 150 nd.

Sociam bicarbonate, NaHCO₄, 10.5 g.

Balance, centigrans (0:01-); precision)

Rallingers, &

Edward of Lakes 125 ml., 6

France !

Charinateri colimier. 25- or 50-rul.

Permanent marker

Provider.

Squatadle.

Whengthing distance, is

Safety Precautions

Acesic acid is a skin and eye irritant. Assis; contact with eyes and skin. West chemical splash griggles, thereical resistant gloves and a chemical resistant sprom. Please consult current Safety Data Showta for additional safety information.

Preparation

- Label etc. Erlemmeyer flacks 1–6. Using a gradianted cylinder, add 25 red, of 1 M aperic acid to each flack.
- 2. Obsain six weighing dishes, and label them 1-6.

 Measure the appropriate ansours of auditors bicarbonate into each weighing dish, according to Table 1.

Table 1.

Saruple	1	2	. 5	4	.5	6
Mass NaHCO _p	0.50 gc	1.00 g	1.50 gr	2.60 gr	2.50 g	3.00 gt

- Obsain etchallorms Stretch the balloons, and blow them up at least once, then let out as much air is possible.
- Use a powder funnel so add sodium bicarbonate sample 1 to one of the halloons.
- b. Platters out the halloon to reviewe any evera sir, then carefully stretch the neck of the halloon over the mouth of Erlenneyer flask 1. Do not allow the solid, to drup into the flask at this time.
- Repeat steps 5 and 6 with excitors bicarbonute samples 2-6.

Procedure

- Introduce the concepts of limiting and evoes reagents by adding stations how many automobiles can be assembled if the following parts are available: 140 car bodies, 520 tree and 270 headlights.
- 2. Which automobile pert is present in a cutority that "limits" the total number of automobiles that may be assembled? In the limiting part the same as the part that is present in the least number? Explain. Note: The tires are the limiting parts in this example, even though there are note tires than anothing else.
- Show stationate the balloon/flack assemblies and ask them to predict what will happen when the scalars bicarbonate is added to the acetic acid in the flack. White the reaction equation on the brand.
- Line up thasks 1—to from right, to befune the leasure desk, Lift each hallows in turn and shake it to allow the solid to fall into the solution. Make sure the neck of the hallows stays firmly attached to the think.
- The reactions will be immediate and rigorous. The white solids will disorder, the solutions will start to buildle and fixe, and the balloom will become inflated.
- 6. Although the relations to princered until the habiliting stops. Compute the stor of the inflated halloons and whether all the wide, has discribed in each case. (The halloon stor should, increase fairly uniformly for flador 1-5, and then stoy constant. It may be hard to tell the differential between flador 4, 5 and 6.).

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