

**Covalent Bonding Quiz (24 Points)****Section I – Basic Concepts & Vocabulary (12 points – 1 pt. each)**

**Instructions:** For Items 1–4, circle the correct response.

1. A *covalent bond* is formed by \_\_\_\_\_ electrons between two nonmetal atoms.  
a) freezing                      c) sharing  
b) transferring                d) melting
2. A group of nonmetal atoms, joined by *covalent bonds* is defined as a(an)  
a) molecule                      c) metal  
b) salt crystal                  d) octet
3. A *structural formula* uses Lewis dot structures to show how atoms are bonded, including the number & types of bonds, in a covalent molecule? (Yes or No)
4. Do atoms, except hydrogen, in covalent bonding follow the octet rule? (Yes or No)

**Instructions:** Match the word on the left with the definition on the right by writing the letter on the line next to the word that is a “definition match” for that word.

- |                            |       |  |       |  |
|----------------------------|-------|--|-------|--|
| 5. molecular formula       | _____ | A. a covalent bond in which only one pair of electrons is shared by two bonded atoms   |       |  |
|                            |       | 6. nonpolar covalent bond  | _____ | B. chemical formula that shows the arrangement of atoms in molecules and polyatomic ions |
| 7. triple covalent bond    | _____ | C. a covalent bond between two atoms of different electronegativities in which the bonding electrons are not shared equally      |       |  |
| 8. double covalent bond    | _____ | D. a covalent bond formed by the equal sharing of bonding electrons by two atoms   |       |  |
| 9. polar covalent bond     | _____ | E. a covalent bond in which two pairs of electrons are shared by two bonded atom   |       |  |
| 10. single covalent bond   | _____ | F. two valence electrons that are <u>not involved in bonding</u> at a bond site in a covalent molecule                           |       |  |
| 11. structural formula     | _____ | G. indicates the exact number of atoms in a covalent compound, but does NOT show the arrangement of the atoms nor types of bonds |       |  |
| 12. unshared electron pair | _____ | H. a covalent bond in which three pairs of electrons are shared by two bonded atom   |       |  |

# Covalent Bonding Quiz With Answers Student Edgenuity

**Rachel Sandford**



## **Covalent Bonding Quiz With Answers Student Edgenuity:**

## Covalent Bonding Quiz With Answers Student Edgenuity Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has been evident than ever. They have the ability to inspire, provoke, and ignite change. Such may be the essence of the book **Covalent Bonding Quiz With Answers Student Edgenuity**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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