

Calculations with Moles:

Using Standard Molar Volume

How many moles of hydrogen are in 100 L of hydrogen at STP?

$$\frac{100 \cancel{\text{L}}}{22.4 \cancel{\text{L}}} \times \frac{1 \text{ mol}}{1} = 4.64 \text{ mol}$$

Calculate Moles Of Elements

Michael Seilmaier

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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