Moles of Elements in a Formula

Subscripts are used to write conversion factors for moles of each element in 1 mole of a compound.

For 1 mole of aspirin, C₉H₈O₄, the possible conversion factors are

$$\frac{9 \text{ moles C}}{1 \text{ mole C,H}_8O_4} = \frac{8 \text{ moles H}}{1 \text{ mole C,H}_8O_4} = \frac{4 \text{ moles O}}{1 \text{ mole C,H}_8O_4}$$

$$\frac{1 \text{ mole C,H}_8O_4}{9 \text{ moles C}} = \frac{1 \text{ mole C,H}_8O_4}{8 \text{ moles H}} = \frac{1 \text{ mole C,H}_8O_4}{4 \text{ moles O}}$$

Calculate Moles Of Elements In A Formula

John McMurry

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