

Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

- 18 mol O₂ 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?



- 21 mol O₂ 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?



- 45 mol H₂O 3. How many moles of water are produced from 22.5 moles of oxygen?



- 20 mol HCl 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.



- 21 mol Cl₂ 5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



Chapter 1 supplemental Problems Stoichiometry Answer Key

Sasha Hu



Chapter 1supplemental Problems Stoichiometry Answer Key:

Stoichiometry and Structure - Freshman Chemistry Problems and how to Solve Them - Part 1 ,1964
Problems Michell J. Sienko,1967

Chemistry

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