

Acids, Bases, and Salts Review

- ① An Arrhenius acid is defined as any compound dissociating in an aqueous solution to form _____ ions. Example: $\text{HNO}_3(\text{aq.}) \rightarrow \text{H}^+(\text{aq.}) + \text{NO}_3^-(\text{aq.})$. Complete the following equation:



- ② An Arrhenius base is defined as any compound dissociating in an aqueous solution to form _____ ions. Example: $\text{KOH}(\text{aq.}) \rightarrow \text{K}^+(\text{aq.}) + \text{OH}^-(\text{aq.})$. Complete the following equation:



- ③ Salts are compounds that dissociate in an aqueous solution releasing neither _____ nor _____ ions. Example: $\text{KCl}(\text{aq.}) \rightarrow \text{K}^+(\text{aq.}) + \text{Cl}^-(\text{aq.})$. Complete the following equation:



- ④ A solution that has strong conductivity and reacts with the metal to make hydrogen gas is a

☐ a Strong Base ☐ b Weak Base ☐ c Strong Acid ☐ d Weak Acid

Answers ____

- ⑤ Which of the following is slippery to touch?

☐ a Strong Acid ☐ b Strong Base ☐ c Strong Electrolyte ☐ d Water

Answers ____

- ⑥ The hydronium ion is

☐ a H_3O^+ ☐ b H_2O ☐ c OH^- ☐ d H_2

Answers ____

- ⑦ A substance that changes color in an acidic or basic solution is called

☐ a Neutralizer ☐ b Inhibitor ☐ c pH indicator ☐ d Catalyst

Answers ____

- ⑧ Name the following acids and bases

HF _____

HNO_3 _____

KOH _____

$\text{Mg}(\text{OH})_2$ _____

Chapter 1review Acids Bases Mixed Answers

Mark Graber, Brigit Ray, Jason K. Wilbur

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Organic Chemistry Michael B. Smith, 2016-03-09 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts *Organic Chemistry An Acid Base Approach* provides a framework for understanding the subject that goes beyond mere memorization Using several techniques to develop a relational understanding it helps students fully grasp the essential concepts at the root of organic chemistry This new edition was rewritten largely with the feedback of students in mind and is also based on the author's classroom experiences using the first edition Highlights of the Second Edition Include Reorganized chapters that improve the presentation of material Coverage of new topics such as green chemistry Adding photographs to the lectures to illustrate and emphasize important concepts A downloadable solutions manual The second edition of *Organic Chemistry An Acid Base Approach* constitutes a significant improvement upon a unique introductory technique to organic chemistry The reactions and mechanisms it covers are the most fundamental concepts in organic chemistry that are applied to industry biological chemistry biochemistry molecular biology and pharmacy Using an illustrated conceptual approach rather than presenting sets of principles and theories to memorize it gives students a more concrete understanding of the material *Modern Chemistry* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

Practical Druggist and Pharmaceutical Review of Reviews, 1925

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