

Introduction to Chemistry

Section 1.1 The Stories of Two Chemicals

In your notebook, read aloud the entire page.

Use each of the terms below just once to complete the passage.

atmosphere
stratosphere

oxygen gas
troposphere

ozone
ultraviolet radiation

ozone hole

Earth's (1) _____ is made up of several layers. The air we breathe makes up the lowest level. This layer is called the (2) _____. The next layer up is called the (3) _____. This level contains a protective (4) _____ layer.

Ozone holes occur when (5) _____ is removed by ultraviolet radiation in the upper part of the atmosphere. The ozone forms a layer around Earth, which absorbs (6) _____. Without ozone, you are more likely to get a sunburn or possibly skin cancer. The thinning of the ozone layer, called the (7) _____, is a problem because without ozone all organisms on Earth are subject to more than one much radiation.

In your notebook, read aloud chlorofluorocarbons.

For each statement below, write true or false.

- _____ (A) CFC is another name for a chlorofluorocarbon.
- _____ (B) CFCs are made up of carbon, fluorine, and oxygen.
- _____ (C) All CFCs are synthetic chemicals.
- _____ (D) CFCs usually react easily with other chemicals.
- _____ (E) CFCs were developed as replacement for toxic refrigerants.

Chapter 1study Guide For Content Mastery Chemistry Answers

**Mary L Kotz,John C. Kotz,William J.
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Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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