

Redox reaction: A chemical reaction in which one substance is oxidized, and another substance is reduced.

Oxidation: when an atom or group of atoms lose electron/electrons, its oxidation is said to take place.

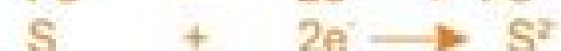
Oxidising agent: The atom or the group of atoms which gain electron/electrons.



Example :



The above reaction is given as



Oxidation of iron: Fe loses 2 electrons

Oxidising agent: S gains 2 electrons

Reduction: when an atom or group of atoms gain electron/electrons, its reduction is said to take place.

Reducing agent: The atom or the group of atoms which lose electron/electrons.



Example :



The above reaction is given as



Reduction of Chlorine: Cl gains 2 electrons

Reducing agent: Na loses 2 electrons

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Yicheng Fang



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