



Reinforcement

Chemical Changes

Read the passage and answer the following questions:



1. What are the products of this chemical reaction?

Two compounds in the reactants become the forming of products in this chemical reaction, the products are water and oxygen. Water is a liquid and oxygen is a gas.

2. What is the reactant in this chemical reaction? Is it a solid, liquid, gas, or a mixture? Is there a symbol for it in the chemical equation?

3. What is the physical state of the reactant and the product?

Hydrogen peroxide is a colorless liquid. It is a mixture of water and oxygen. The products are water and oxygen. Water is a liquid and oxygen is a gas. The products are water and oxygen. The products are water and oxygen.

4. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much oxygen is formed? How much water is formed?

100 grams



5. What is the chemical formula for the product? Is it a solid, liquid, gas, or a mixture?

6. What is the chemical formula for the reactant? Is it a solid, liquid, gas, or a mixture?

7. What are the physical states of the reactant and the product?

Hydrogen peroxide is a liquid.
Water is a liquid and oxygen is a gas.

8. What is the physical state of the product? Is it a solid, liquid, gas, or a mixture?

9. What is the chemical formula for the product? Is it a solid, liquid, gas, or a mixture?

Hydrogen peroxide is a liquid. Water is a liquid and oxygen is a gas.

10. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much oxygen is formed? How much water is formed?

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McGraw-Hill Staff



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