



Problem Set 4
Chapters 4 and 25

Due Thursday Oct 24th

Nuclear Chemistry

Essay Due Oct 30th

Test Wednesday October 30th

- Your work is to be done on separate sheets of paper.
- All of the papers should be stapled together in the upper left hand corner with this question sheet at the back of the packet.
- Each of your responses should be numbered the same way as it appears on this packet.

Conceptual Questions

- 1) State the law of conservation of mass.
- 2) Why are atoms unstable?
- 3) How do unstable atoms gain stability?
- 4) A radioactive element having atomic number 82 and atomic mass 214 loses a beta particle, $0, -1\beta$. What is the resulting particle?
- 5) What type of decay is the reaction below, and what isotope completes this nuclear equation?
 $27,13\text{Al} + 2,1\text{H} \rightarrow 4,2\text{He} + ?$

Practice Questions

- 6) Complete this chart on the three types of radioactive decay

| Type | Symbol | Mass (amu) | Charge |
|-------|--------|------------|--------|
| Alpha | | | |
| Beta | | | |
| Gamma | | | |

- 7) If the radioactive atom $238, 92\text{U}$ emits an alpha particle, the atom remaining is represented by
 (A) $238, 92\text{U}$ (B) $236, 92\text{Th}$ (C) $234, 92\text{U}$ (D) $234, 90\text{Th}$ (E) $234, 91\text{Pa}$
- 8) What term completes this equation? $30,14\text{Si} + ? \rightarrow 32,16\text{S} + 0, -1e$
 (A) $4,2\text{He}$ (B) $2,1\text{H}$ (C) $1,1\text{H}$ (D) $0, -1e$
- 9) What particle is emitted when lead-214, $214, 82\text{Pb}$, undergoes natural radioactive decay to bismuth-214, $214, 83\text{Bi}$?
 (A) an alpha particle, $4,2\text{He}$ (C) a proton, $1,1\text{H}$
 (B) a beta particle, $0, -1e$ (D) a neutron, $1,0\text{n}$

Extension Question

- 10) Recently Alexander Valterovich Litvinenko was poisoned by radioactive polonium-210.
 - a. Research who he was and why he was poisoned.
 - b. Find out what polonium-210 does to people.

Chapter 25 Nuclear Chemistry Worksheet Answers

Peipei Pang



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Table of Contents Chapter 25 Nuclear Chemistry Worksheet Answers

1. Understanding the eBook Chapter 25 Nuclear Chemistry Worksheet Answers
 - The Rise of Digital Reading Chapter 25 Nuclear Chemistry Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 25 Nuclear Chemistry Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 25 Nuclear Chemistry Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 25 Nuclear Chemistry Worksheet Answers
 - Personalized Recommendations
 - Chapter 25 Nuclear Chemistry Worksheet Answers User Reviews and Ratings
 - Chapter 25 Nuclear Chemistry Worksheet Answers and Bestseller Lists
5. Accessing Chapter 25 Nuclear Chemistry Worksheet Answers Free and Paid eBooks
 - Chapter 25 Nuclear Chemistry Worksheet Answers Public Domain eBooks
 - Chapter 25 Nuclear Chemistry Worksheet Answers eBook Subscription Services
 - Chapter 25 Nuclear Chemistry Worksheet Answers Budget-Friendly Options

6. Navigating Chapter 25 Nuclear Chemistry Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 25 Nuclear Chemistry Worksheet Answers Compatibility with Devices
 - Chapter 25 Nuclear Chemistry Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 25 Nuclear Chemistry Worksheet Answers
 - Highlighting and Note-Taking Chapter 25 Nuclear Chemistry Worksheet Answers
 - Interactive Elements Chapter 25 Nuclear Chemistry Worksheet Answers
8. Staying Engaged with Chapter 25 Nuclear Chemistry Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 25 Nuclear Chemistry Worksheet Answers
9. Balancing eBooks and Physical Books Chapter 25 Nuclear Chemistry Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 25 Nuclear Chemistry Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 25 Nuclear Chemistry Worksheet Answers
 - Setting Reading Goals Chapter 25 Nuclear Chemistry Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 25 Nuclear Chemistry Worksheet Answers
 - Fact-Checking eBook Content of Chapter 25 Nuclear Chemistry Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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