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## Chapter 2: Internal Energy (U), Work (w), Heat (q), Enthalpy (H)

Internal Energy (includes motion and rotation of vessel)

→ look at isolated part of universe



$$U^* = U_{\text{cylinder}} + U_{\text{piston}}$$

Universe = isolated

First Law of Thermodynamics:

- Total  $U^*$  for isolated system is constant
- Energy can be exchanged between various subsystems
- Energy forms can be interconverted  
Eg. Chemical En. → Heat → Work

$$\Delta U_{\text{cylinder}} = \Delta U_{\text{cylinder}} + \Delta U_{\text{piston}} = 0$$

# Chapter 2 Heat Work Internal Energy Answers

**Wilbur Olin Atwater, Francis Gano  
Benedict**



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