

## Section 6-1

## Concept Review

## Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 10 neutrons, and 12 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a octet.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

noble gases are stable while everything  
else is missing some

# Concept Review Section Simple Ions Answers

**David M. Hillis**



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## **Table of Contents Concept Review Section Simple Ions Answers**

1. Understanding the eBook Concept Review Section Simple Ions Answers
  - The Rise of Digital Reading Concept Review Section Simple Ions Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Concept Review Section Simple Ions Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Concept Review Section Simple Ions Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Concept Review Section Simple Ions Answers
  - Personalized Recommendations
  - Concept Review Section Simple Ions Answers User Reviews and Ratings

- Concept Review Section Simple Ions Answers and Bestseller Lists
- 5. Accessing Concept Review Section Simple Ions Answers Free and Paid eBooks
  - Concept Review Section Simple Ions Answers Public Domain eBooks
  - Concept Review Section Simple Ions Answers eBook Subscription Services
  - Concept Review Section Simple Ions Answers Budget-Friendly Options
- 6. Navigating Concept Review Section Simple Ions Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Concept Review Section Simple Ions Answers Compatibility with Devices
  - Concept Review Section Simple Ions Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Concept Review Section Simple Ions Answers
  - Highlighting and Note-Taking Concept Review Section Simple Ions Answers
  - Interactive Elements Concept Review Section Simple Ions Answers
- 8. Staying Engaged with Concept Review Section Simple Ions Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Concept Review Section Simple Ions Answers
- 9. Balancing eBooks and Physical Books Concept Review Section Simple Ions Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Concept Review Section Simple Ions Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Concept Review Section Simple Ions Answers
  - Setting Reading Goals Concept Review Section Simple Ions Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Concept Review Section Simple Ions Answers
  - Fact-Checking eBook Content of Concept Review Section Simple Ions Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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