

Now _____ Date _____ Class _____

CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

Covalent Bonding

Section 9.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	p bond
---------------	----------	------------	------------	--------

When sharing of electron occurs, the attachment between atoms that results is called a(n) **(1) covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **(2) exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **(3) molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **(4) sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **(5) p bond**.

In your textbook, read about single and multiple bonds and bond strengths.

Circle the letter of the choice that best completes the statement or answers the question.

- Is what form the elements such as hydrogen, nitrogen, and oxygen normally exist?
 - as single atoms
 - ☒ as molecules containing three atoms
 - as molecules containing two atoms
 - as molecules containing four atoms
- How many electrons are shared in a double covalent bond?
 - none
 - one
 - two
 - four
- Bond length is the distance between
 - two molecules of the same substance
 - ☒ the nuclei of two attached atoms
 - the electrons in two attached atoms
 - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
 - ☒ the shorter the bond, the stronger the bond
 - the shorter the bond, the weaker the bond
 - the shorter the bond, the lower the electron in it
 - the shorter the bond, the lower the bond dissociation energy

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CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

Section 9.2 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write true or false.

- Binary molecular compounds are generally composed of a metal and a nonmetal. **false**
- The second element in the formula of a binary compound is named using the suffix -ite. **false**
- The prefix zero indicates three atoms. **false**
- The prefix four indicates six atoms. **true**
- In naming the first element in a formula, the prefix zero is not used. **true**
- For binary acids, the hydrogen part of the compound is named using the prefix hydro. **true**
- An oxyacid contains only two elements. **false**
- If the name of the anion of an oxyacid ends in -ite, the acid name contains the suffix -ous. **false**

In your textbook, read about naming molecular compounds and oxyacids.

For each item in Column A, write the letter of the matching item in Column B.

Column A

- C_2H_4
- H_2CO_3
- H_2CO
- $\text{H}_2\text{C}_2\text{O}_4$
- H_2SO_4
- H_2SO_3
- H_2CO_3
- H_2SO_4
- H_2SO_3
- H_2CO_3
- H_2SO_4
- H_2SO_3
- H_2CO_3
- H_2SO_4
- H_2SO_3

Column B

- hydrobromic acid
- nitrogen tetroxide
- carbon monoxide
- nitrous acid
- acetic acid
- nitric acid
- carbonic acid
- nitrous acid
- carbon dioxide

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Chapter 1stoichiometry Study Guide For Content Mastery Key

**Denise Guinn, Rebecca Brewer, Rachel
C. Lum**



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This book delves into Chapter 1stoichiometry Study Guide For Content Mastery Key. Chapter 1stoichiometry Study Guide For Content Mastery Key is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chapter 1stoichiometry Study Guide For Content Mastery Key, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 5: Conclusion
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 6. In chapter 5, the author will draw a conclusion about Chapter 1stoichiometry Study Guide For Content Mastery Key. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chapter 1stoichiometry Study Guide For Content Mastery Key.

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