## Stoichiometry Problem Set A

Directions: Solve each of the following problems. Show work using the factor label method or the irf box. Include correct units and significant figures with each answer. Clearly indicate final answer. Be sure to start with a balanced equation

#### Part 1: Straight Stoichiometry Problems

Target Goal: To properly calculate the amount of one substance equal to a given amount of another substance within a chemical reaction using the factor-label method.

For the reaction HNO<sub>Neq</sub> + Mg(OH)<sub>100</sub> → Mg(NO<sub>2</sub>)<sub>3(eq)</sub> + H<sub>2</sub>O<sub>(0)</sub>, how many grams of magnesium nitrate are produced from 8.00 mol of nitric acid (HNO<sub>2</sub>)? A

MM=148.39

For the reaction 2 KClO<sub>300</sub> → 2 KCl<sub>00</sub> + 3 O<sub>200</sub>, how many moles of potassium chlorate are required to produce 160. g of oxygen? A

MH = 32.09

- KCID3
- 3.) For the reaction  $2H_{200} + O_{200} \rightarrow 2H_2O_{50}$ , how many grams of water are produces from 5.22 liters of oxygen gas?

MH= 18.09

4.) In the reaction Na<sub>(0)</sub> + PH<sub>2</sub>O<sub>(0)</sub> → NaOH<sub>0-0</sub> + H<sub>2(0)</sub>, how many grams of hydrogen are produced if 1.33 grams of sodium react completely?

MH= 23.09

5.) In the reaction  $CO_{200} + H_2O_{00} \Rightarrow H_2O_{200}$ , how many moles of carbonic acid can be produced from 97.2 g of  $CO_2$ ?

MH 44.09

2.21 moles Heca

#### Did you achieve our target goal?

Target Goal: To properly calculate the amount of one substance equal to a given amount of another substance within a chemical reaction using the factor-label method. Circle your answer.

Feeling VERY confused

3 I think I am catching on

Doing great! Bring it on!

# **Chapter 1supplemental Problems Stoichiometry Answers**

**IM Harris** 

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Chemistry McGraw-Hill Staff,2001-03 Resources in Education ,1991-04 **Chemical Problem Solving Using Dimensional Analysis** Robert Nakon, 1978 **Chemistry** John S. Phillips, Cheryl Wistrom, 2000 Holt Chemistry Ralph Thomas Myers, 2004 Manual Nitrogen Control ,1993 Item no 0431 K Scientific and Technical Aerospace Reports .1984 Basic Calculations for Chemical and Biological Analysis Bassey J. S. Efiok, Etim Effiong Eduok, 2000 Like the 1993 edition this iteration does not assume that students lab technicians and scientists have mastered the prerequisite calculation skills for quantitative problems in the chemical biomedical sciences A new chapter focuses on using spreadsheets and laboratory information management systems Other chapters cover calculations and techniques relevant to reagents chemical reactions properties of gases and solutions pH and buffer preparation spectrophotometry enzyme assays and radioactivity Also included are derivations of some key equations quick reference guides and an index to the practical examples Efiok is with the National Heart Lung and Blood Institute National Institutes of Health Eduok is in the chemistry department at Xavier U of Louisiana c Book News Inc Chemical Engineering Education ,1980 **Abstracts of Papers -**Resources in American Chemical Society American Chemical Society, 1975 **Government Reports Announcements & Index**, 1995-03 Education ,1982 Nuclear Energy ,1996 **Government Reports Annual Index** ,1992 The Energy Index ,1977 **Book of Abstracts** American Chemical Bibliography of Agriculture ,1984 Society, Meeting, American Chemical Society, 1976 Bibliography of Agriculture with Subject Index ,1984 Supplemental Problems - Glencoe Chemistry Matter and Change, 2013 **Elementary** Chemistry Keith S. Anliker, 2007-08-20

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