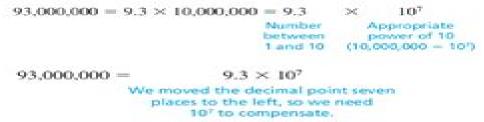
Chapter 2: Measurements and Calculations

- 2.1 Scientific Notation
 - A quantitative observation is a measurement using numbers
 - Example: a substance weighing 4.3 grams
 - Numbers associated with scientific measurements are often large or very small so they are expressed with scientific notation
 - Scientific notation expresses a number as a product of a number between 1 and 10 and the appropriate power of 10.



2.2 Units

- A Unit is a measurement that tells us the scale/standard that is being used to represent the result of the of measurement
- Two most widely used systems are the English system used in the United States and the metric system used in most of the rest of the industrialized world
- In 1960, the International System was set up which was a comprehensive system of units.

	Table	2.1	Some	Fundamental	SI Units
--	-------	-----	------	--------------------	----------

Physical Quantity	Name of Unit	Abbreviation
mass	kilogram	kg
length	meter	m
time	second	s
temperature	kelvin	K

- Since fundamental units are not always a convenient size, the International System or SI system uses prefixes to change the size of the unit
- Measurements consist of both a number and a unit
- 2.3 Measurements of Length, Volume, and Mass
 - Fundamental SI unit of length is meter (1 meter = 39.37 inches)
 - Fractions of a meter or multiples of a meter can be expressed by powers of 10
 - Volume is the amount of a 3D space occupied by a substance
 - Fundamental unit of volume in the SI is based on a cube that measures 1 meter in each
 of the three directions

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