Acids and Bases Problems

Rank the following solutions in order of increasing acidity (from lowest to highest):

Solution A with pH = 6.50 Solution B with $[H_3O^*] = 3.5x10^{-5} M$ Solution D with pH = 5.85

Solution C with [OH-] = 6.0x10-10 M

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2. Complete the following table.

Acid	Conjugate Base
HS-	
	CIO ₄ -
H ₃ PO ₄	
	HCIO ₃ -

3 Write the acid-base reactions with water for each of the following acids. Label the acid, base, conjugate acid, and conjugate base.

i) [Co(NH₃)₅(OH₃)]3*

HSO,

iii) CH₂OH

W) H,PO,

Chapter 2acids Bases Practice Problems Answers

Victor M. Corman

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