

Acids Bases and Salts Class 10 Important Questions with Answers Science Chapter 2

July 5, 2022 by Bhagya

We have given these [Important Questions for Class 10 Science](#) Chapter 2 Acids Bases and Salts to solve different types of questions in the exam. Previous Year Questions & Important Questions of Acids Bases and Salts Class 10 Science Chapter 2 will help the students to score good marks in the board examination.

Important Questions of Acids Bases and Salts Class 10 Science Chapter 2

Question 1.

With the help of an example explain what happens when a base reacts with a non-metallic oxide. What do you infer about the nature of non-metal oxide? (Board Term I, 2017)

Answer:

Oxides of non-metals react with bases to form salt and water. For example, the reaction between carbon dioxide and calcium hydroxide. Calcium hydroxide, which is a base, reacts with carbon dioxide to produce salt and water.



Hence, oxides of non-metals are acidic in nature.

Question 2.

What is observed when carbon dioxide gas is passed through lime water

(i) for a short duration?

(ii) for a long duration? Also write the chemical equations for the reactions involved. (Board Term I, 2016)

Answer:

(i) When CO₂ is passed through lime water for short interval of time, it turns milky due to the formation of insoluble calcium carbonate.



(ii) If CO₂ is passed for long duration through lime water, the white precipitate formed dissolves due to the formation of soluble calcium hydrogen carbonate and the solution becomes clear.



Question 3.

2 mL of sodium hydroxide solution is added to a few pieces of granulated zinc metal taken in a test tube. When the content are warmed, a gas evolves which is bubbled through a

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William H. Brown, Thomas Poon



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