## Science

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(Chapter - 1) (Matter in our Surroundings) (Exercises) (Class - IX)

#### Question 1:

Convert the following temperatures to the Celsius scale.

(a) 293 K

(b) 470 K

Answer 1:

- (a)  $293 \text{ K} = (293 273)^{\circ}\text{C} = 20 ^{\circ}\text{C}$
- (b)  $470 \text{ K} = (470 273)^{\circ}\text{C} = 197 ^{\circ}\text{C}$

#### Ouestion 2:

Convert the following temperatures into the Kelvin scale.

(a) 25°C

(b) 373 °C

- \_Answer 2:
- (a) 25 °C = (25 + 273) K = 298 K
- (b)  $373 \,^{\circ}\text{C} = (373 + 273) \,^{\circ}\text{K} = 646 \,^{\circ}\text{K}$

#### Ouestion 3:

Give reasons for the following observations.

- (a) Naphthalene balls disappear with time without leaving any solid.
- (b) We can get the smell of perfume sitting several meters away.
- \_Answer 3:

(a) Naphthalene shows the property of sublimation. Evaporation of naphthalene takes place easily and so it disappears during course of time without leaving a solid.

(b) Perfumes vaporize very fast and its vapours diffuse into air easily. That is why we can smell perfume sitting several meters away.

#### Outestion 4:

Arrange the following substances in increasing order of forces of attraction between the particles water, sugar, oxygen.

#### Answer 4:

Oxygen < Water < Sugar.

#### Ouestion 5:

What is the physical state of water at -

(a) 25°C

(b) 0°C

(c) 100°C?

- Answer 5:
- (a) Liquid
- (b) Solid and Liquid
- (c) Liquid and Vapours

#### Question 6:

Give two reasons to justify:

- (a) Water at room temperature is a liquid.
- (b) An iron almirah is a solid at room temperature.

Answer 6:

(a) Water at room temperature is a liquid because it has fluidity and has definite volume but no definite

(b) An iron almirah is a solid at room temperature because it is rigid and has a definite shape.

# **Chapter 1science Test Answers**

**Caren Van Slyke** 

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